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1	Stable isotope geochemistry of silicon in granitoid zircon
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16	Keywords: Si isotopes, zircon, granite, LA-MC-ICP-MS
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19	ABSTRACT
20	Zircon is often used to study granites (sensu lato) and continental crust because it
21	is very resistant and can be analyzed for various isotope systems that provide time
22	and source information about their parental melts. Granites with different
23	petrological histories have distinct bulk-rock silicon isotope compositions but it is
24	unclear if these differences are also detectable in zircon because of superimposed
25	fractionation effects (e.g., related to temperature, silica content, magmatic
26	processes). The present study explores the Si isotope signatures of zircon from
27	various granite types to constrain their isotope fractionation behavior and uses

28 them as igneous petrogenetic tools, and possibly as granite source discriminators 29 when zircon is found in detrital sediments. Our results show that although Si isotope compositions in zircon can be modified by secondary (post-crystallization) 30 31 processes such as alteration/weathering and metamorphism, they are primarily 32 controlled by zircon-melt isotope fractionation, which depends on both zircon crystallization temperature and magma silica content. Once these fractionation 33 34 effects are understood and filtered out, a pattern emerges between Si isotope 35 signatures of zircons from different granite types that is consistent with theoretical 36 and experimental results as well as with known Si isotope differences at the bulk-37 rock scale. Silicon isotope ratios in zircon can track magma evolution (e.g., temperature and SiO₂ changes) and, hence, reveal complex processes that involved 38 39 magma mingling, fractional crystallization, and/or multiple sources. This study, 40 therefore, illustrates that Si isotopes in zircon can be used to investigate magma evolution and represents a useful complement to existing techniques in granite 41 42 studies involving zircon (e.g., U-Th-Pb, Lu-Hf and O isotopes) provided that it is 43 not used as a stand-alone technique.

44

45 **1. Introduction**

Silicon is a major component of silicates which make up most of the terrestrial mantle and crust. This chemical element has three stable isotopes 28, 29, and 30, with respective mean abundances of 92.23%, 4.68%, and 3.09% (Debievre and Taylor, 1993). Although developed long ago (e.g., Reynolds and Verhoogen, 1953; Clayton, 1978; Douthitt, 1982), Si isotope measurements have been applied to high-temperature geochemistry only recently (e.g., Shahar et al., 2009; Savage et al., 2011, Zambardi and Poitrasson, 2011) thanks to the advent of multicollection inductively coupled plasma 53 mass spectrometry (MC-ICP-MS; e.g., Georg et al., 2006) and optimized sample 54 preparation procedure (i.e., alkali fusion) that allowed sufficiently high precision to 55 observe natural variations. High-temperature Si isotope fractionation has consequently 56 been studied in various contexts from cosmochemistry to igneous petrology allowing 57 the establishment of bulk silicate Earth composition, which is distinct from most other 58 planetary bodies (e.g., Georg et al., 2007; Savage et al., 2010; Shahar et al., 2011; 59 Ziegler et al., 2012; Zambardi et al., 2013; Moynier et al., 2020), and the identification 60 of a general igneous differentiation trend towards isotopically heavy Si isotope 61 signatures for SiO₂-rich magmas (e.g., Savage et al., 2011; Poitrasson, 2017). In 62 addition, it has been shown recently that granitoids - plutonic rocks formed by the slow 63 cooling of a silica-rich magmas – exhibit distinct silicon isotope signatures depending 64 on their chemical affinity and, hence, their source (e.g., Savage et al., 2012; André et 65 al., 2019; Deng et al., 2019; Fig. 1). There are several ways to produce granitic magmas 66 which is reflected in their chemical and isotopic compositions, ultimately responsible, 67 together with pressure and temperature of crystallization, for the mineralogy of these plutonic rocks (e.g., Nédélec et al., 2015). Granites have been grouped into multiple 68 69 types according to various classifications (e.g., peraluminous, metaluminous: Shand, 70 1943; A, I, S: Chappell and White, 1974; MPG, CPG, KCG, ACG, RTG, PAG: 71 Barbarin, 1999; Table 1) essentially based on petrological and/or geochemical criteria. 72 Granites, and more generally granitoids, are the major constituents of continents and it 73 is, therefore, important to understand their formation and evolution over time because 74 continents witnessed almost the entire Earth's evolutionary history.

Granites (sensu lato) most commonly contain zircon (ZrSiO₄) but in variable
quantities depending on their origin and their degree of differentiation (e.g., Pupin,
1980; Samson et al., 2018). Zircon can be precisely and accurately dated using U-Th-

78 Pb isotope systems (e.g., Schoene, 2014) while the source of its parental magma can be 79 approached using Hf and O isotopes (e.g., Patchett, 1983; Valley, 2003; Payne et al., 80 2016). Moreover, zircon is very resistant to weathering and erosion, and most elements 81 diffuse very slowly within its lattice (e.g., Lee et al., 1997; Cherniak and Watson, 2003) 82 making this mineral a very efficient time-capsule. Therefore, zircon has been a very 83 useful source of information for granite petrogenesis. It is currently used in magmatic rocks and in detrital sediments, as an archive of the crust, to reconstruct the formation 84 85 and evolution of continents (e.g., Roberts and Spencer, 2015 and references therein). 86 Isotopic information provided by zircon should be interpreted differently for each 87 granite type because of the distinct sources and petrogenetic processes involved in their 88 formation (e.g., Barbarin, 1999; Bonin et al., 2019; Moyen, 2019). While only a few 89 tools exist to determine the nature of the parental rock of a detrital zircon, they remain 90 ambiguous and not always reliable (e.g., Belousova et al., 2006). Since granites of 91 different types (A, I, S, and TTG) can have distinct Si isotope signatures (Savage et al., 92 2012; Poitrasson and Zambardi, 2015; André et al., 2019; Deng et al., 2019), it is likely 93 that these differences can be retrieved from the Si isotope signature of zircon 94 crystallized in these rocks.

95 The silicon stable isotope signature of an igneous rock can be controlled by that 96 of its source, mineralogy, and melting/crystallization history (Savage et al., 2011; 97 Poitrasson, 2017). In contrast, the Si isotope signature of a zircon is controlled by the 98 degree of fractionation with its parental magma which essentially depends on the 99 temperature of crystallization and the silica content (~degree of polymerization) of the 100 surrounding melt that was studied theoretically and experimentally by Qin et al. (2016) 101 and Trail et al. (2019). Co-precipitating phases (e.g., feldspar, quartz) can also play a 102 role provided that they formed in equilibrium. However, post-crystallization processes

103 (e.g., radiation-damage, metamorphic recrystallization, alteration and weathering; Ewing et al., 2003; Vavra et al., 1999; Pidgeon et al., 2017, 2019) could possibly modify 104 105 the Si isotope composition of a zircon. Trail et al. (2018) and Chowdhury et al. (2020) 106 already presented Si isotope data for natural zircons and illustrated the potential of such application for very old igneous rocks. However, we thought it was appropriate to 107 108 present a more extensive dataset of natural zircons from a wider variety of granite types 109 to more deeply understand the fractionation behavior between zircon and its parental-110 rock and, hence, complement theoretical (Qin et al., 2016) and experimental (Trail et 111 al., 2019) constraints. This, in turn, would provide a more robust reference frame for 112 the interpretation of Si isotope data in zircon, and illustrate whether or not they can be 113 used to discriminate granite types. Moreover, adding Si isotopes to the isotope tool-box 114 currently applied to detrital zircons would help clarify the detrital record by better 115 constraining the nature of zircon parental rocks and, therefore, prevent ad hoc 116 interpretations of O and Hf isotope signatures.

117 In order to address these issues, this article presents Si isotope data obtained by laser-ablation multicollection inductively coupled plasma mass spectrometry (LA-MC-118 119 ICP-MS) and solution multicollection inductively coupled plasma mass spectrometry 120 (S-MC-ICP-MS) for zircons from various granite types (4 A-type, 9 tonalitetrondhjemite-granodiorite: TTGs, 10 I-type, and 3 S-type; Tables 1 and S1), together 121 122 with LA-ICP-MS trace element concentrations, U-Pb ages, and LA-MC-ICP-MS Lu-123 Hf isotope data for a subset of these zircons. We acknowledge that the A-I-S 124 classification does not encompass the entire diversity of granites (e.g., Barbarin, 1996, 125 1999; Bonin et al., 2019; Moyen et al., 2016; Moyen, 2019) but have decided to use it 126 in the present study because it has the advantage of being widely used. Our new results show that Si isotopes in zircon display a fractionation behavior that complies very well 127

with theory and experiment but, as suspected, secondary processes (e.g., weathering and metamorphism) can compromise the Si isotope signature of zircon. However, trace element signatures can help filter out altered compositions. Once fractionation behavior is understood and alteration filtered out, zircon Si isotope signatures are consistent with those in bulk-rocks.

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134 **2. Material and Methods**

135 **2.1. Samples**

136 Zircon crystals studied here are from felsic igneous rocks of various affinities (e.g., A, I, S, TTG; Table 1) and correspond to separate minerals mounted in epoxy. The A-type 137 138 granites analyzed in this study are from the Pikes Peak batholith, the Colorado Front 139 Range (USA; Guitreau et al., 2016) and the Skjoldungen complex (Greenland; Nutman 140 and Rosing, 1994) but we also analyzed a detrital zircon megacryst (1.5 cm long) from 141 Tambani (Malawi) that derives from nepheline-bearing gneiss (Ashwal et al., 2007). 142 Granites of I-type affinity analyzed here are from the Skellefte district (Sweden; 143 Guitreau et al., 2014a and references therein), the Acasta Gneiss Complex (Canada; 144 Guitreau et al., 2014b), and the French Massif Central (France; Barbarin, 1988a). Note 145 that some of the Skellefte samples are not granites but sub-volcanic or porphyritic eruptive products that are chemically akin to and coeval with I-type granites with which 146 147 they are associated. S-type granites are from the French Massif Central (Barbarin, 148 1988b) and correspond to cordierite granite (CPG) and two-mica granite (MPG; 149 Barbarin, 1999). TTG studied here are from the Acasta Gneiss Complex (Canada), Barberton Greenstone Belt (South Africa), the Karelian craton (Finland), the 150 151 Sharyzalgai uplift (Siberia), and the Napier Complex (East Antarctica). More information about these samples can be found in Guitreau et al. (2012) and referencestherein.

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2.2. Zircon imaging and analysis

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2.2.1. Zircon CL and BSE imaging

157 Zircon cathodoluminescence (CL) and back-scattered electron (BSE) imaging was done at Laboratoire Magmas et Volcans (LMV, France) using a Jeol JSM-5910LV 158 159 scanning electron microscope. Image acquisition was done using an acceleration 160 voltage of 15 kV. Zircon imaging revealed internal textures which served as a tool to interpret zircon crystallization conditions (e.g., mafic or felsic igneous, metamorphic) 161 162 as well as radiation damage accumulation and alteration effects (e.g., Corfu et al., 163 2003). This ultimately guided elemental and isotopic measurements that were located 164 as much as possible in igneous domains (based on texture), except in the case of the zircon megacryst from Malawi in which domains corresponding to various 165 166 crystallization conditions were purposely analyzed. Zircon crystals from TTGs and 167 some I- and A-type granites were already imaged using CL/BSE technique and analyzed for Lu-Hf isotopes and U-Pb ages (Guitreau et al., 2012, 2014a, 2016, 2018, 168 169 2019).

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2.2.2. Si isotope measurements by LA-MC-ICP-MS

172 Laser-ablation multicollection inductively coupled plasma mass spectrometry 173 measurements were conducted at LMV using a Resonetics Resolution M-50E coupled 174 to a Thermo Scientific Neptune Plus MC-ICP-MS, and followed closely the analytical 175 protocol of Guitreau et al. (2020). Analytical conditions are provided in Table 2. Si 176 isotopes were measured as traverses and the international zircon standard 91500 (from 177 a syenite; Wiedenbeck et al., 1995) was used as a bracketing standard. The zircon 178 standards MudTank (from a carbonatite; Black and Gulson, 1978), KIM (from a 179 kimberlite; Trail et al., 2018), AS3 (from a gabbro; Paces and Miller, 1993), Plešovice 180 (hyperpotassic granulite; Sláma et al., 2008), and the synthetic standards MUN from the Memorial University of NewFoundland (Fisher et al., 2011) were used for quality 181 check. Silicon isotope results are reported in Tables S2, S3, and 3 in delta notation, 182 δ^{29} Si and δ^{30} Si, which express per mil deviations from a reference (quartz NBS 28), 183 184 and are calculated using the following equations.

185
$$\delta^{X} Si = \left[\frac{(XSi)^{28}Si_{sample}}{(XSi)^{28}Si_{standard}} - 1 \right] \times 1000$$

186 where X is either 29 or 30.

187 Throughout the manuscript, analytical uncertainty (i.e., within-run precision) is 188 reported as 2 standard-errors (2SE) whereas external reproducibility (e.g. average of 189 multiple analyses of the same zircon) is reported as 2 standard-deviations (2SD).

Silicon isotope data for AS3 zircon analyzed by LA-MC-ICP-MS in this study 190 gave a δ^{30} Si of $-0.47 \pm 0.20\%$ (2SD; Table 3), coherent with the value of $-0.36 \pm 0.35\%$ 191 192 (2SD obtained by Trail et al. (2018) using SIMS and the value of $-0.44 \pm 0.23\%$ (2SD) obtained by Guitreau et al. (2020) using the same LA-MC-ICP-MS set-up as the present 193 study. KIM zircon returned a LA-MC-ICP-MS $\delta^{30}Si$ of -0.41 \pm 0.19‰ which is 194 identical to the S-MC-ICP-MS value of $-0.41 \pm 0.01\%$ (Trail et al. 2018) and the LA-195 MC-ICP-MS δ^{30} Si of -0.40 ± 0.23‰ from Guitreau et al. (2020). MudTank zircon 196 analyzed in this study exhibit a LA-MC-ICP-MS δ^{30} Si of -0.38 ± 0.23‰ which is 197 198 consistent, within error, with the S-MC-ICP-MS value of $-0.34 \pm 0.03\%$ from Trail et al. (2018) and the δ^{30} Si of -0.47 ± 0.17% obtained by Guitreau et al. (2020). Plešovice 199 zircon gave a δ^{30} Si of -0.44 ± 0.21‰ which is consistent with S-MC-ICP-MS data of -200 201 $0.39 \pm 0.06\%$ from Guitreau et al. (2020). The synthetic MUN zircons analyzed in this

study exhibited δ^{30} Si of -1.61 ± 0.21‰ for MUN-0 and -2.02 ± 0.27‰ for MUN-1, which agree with values of -1.67 ± 0.14‰ and -2.08 ± 0.17‰, respectively, reported in Guitreau et al. (2020).

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2.2.3. Si isotope measurements by S-MC-ICP-MS

207 We analyzed some of the zircons by S-MC-ICP-MS for the sake of comparison with LA-MC-ICP-MS measurements. Solution-MC-ICP-MS measurements were conducted 208 during the same analytical sessions as zircon standard data from Guitreau et al. (2020) 209 210 which used a technique modified after (Deng et al., 2019). Results for basalt reference material BHVO-2 normalized to quartz reference NBS28 analyzed using MC-ICP-MS 211 alongside unknowns comply well with consensus values as we obtained δ^{29} Si and δ^{30} Si 212 values of -0.14 \pm 0.09‰ and -0.28 \pm 0.15‰ (n = 60; 2SD) for BHVO-2 (δ^{30} Si = -0.29 213 $\pm 0.05\%$; Schuessler and Von Blackenburg, 2014). 214

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216 **2.2.4.** Trace element measurements in zircon

217 Trace element measurements were done at LMV by LA-ICP-MS using the same laser 218 as for Si isotopes but coupled to a Thermo Scientific Element XR ICP-MS. Analytical 219 spots were located on shallow lines formed by Si isotope analyses. Information regarding analytical conditions are given in Table 2. We used ²⁸Si as an internal 220 standard and the synthetic glass standard SRM NIST 610 (Jochum et al., 2011) as a 221 222 bracketing standard, whereas SRM NIST 612 (Jochum et al., 2011), 91500 223 (Wiedenbeck et al., 1995, 2004), MudTank (Gain et al., 2019), and AS3 (Takehara et al., 2018) zircon standards were analyzed for quality check. Batches of 6 unknowns and 224 225 3 quality-control standards were bracketed by 2 analyses of NIST 610. We used published Si concentration data when available but, otherwise, used stochiometric 226

227 values to convert ICP-MS counts/second into concentrations. Results are presented in Table S4. Elemental concentrations obtained for SRM NIST 612 were all consistent 228 with published values within $\leq 7\%$ (Jochum et al., 2011). Data for zircon standards 229 230 analyzed in this study complied well with available published values for most elements 231 (Table S4; Fig. S1). Note, however, that some elemental concentrations fall within the 232 range of published values or close in the case of zircon reference materials that exhibit 233 compositional differences between batches (MudTank and AS3) and/or because of 234 alteration (e.g., Ca, Al, K, La contents).

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2.2.5. U-Pb geochronology in Malawi zircon

237 The megacrystic zircon from Malawi was the only crystal studied for U-Pb and Lu-Hf 238 in this work because its size allowed spatial resolution on various domains such as magmatic, metamorphic, and altered ones, and we wanted to investigate possible 239 240 relationship between geochemical signatures within these domains. The regular size of 241 normal zircons (~100-200 µm) does not allow these domains to be individualized. 242 Therefore, this zircon was the appropriate candidate to test the effect of magmatic and 243 metamorphic growth as well as secondary alteration on the geochemistry of zircon. In 244 addition, a fair number of other zircons studied in this study were previously analyzed 245 for U-Pb and Lu-Hf isotope systematics in earlier studies (Guitreau et al., 2012, 2014b, 246 2019). U-Th-Pb isotope measurements were done at LMV by LA-ICP-MS using the 247 same instruments as for trace elements. Analytical spots were located next to lines formed during Si isotope analyses. Information regarding analytical conditions are 248 249 given in Table 2. The zircon standard 91500 (1065.4 \pm 0.3 Ma; Wiedenbeck et al., 250 1995) was used for bracketing whereas AS3 (1099.1 \pm 0.5 Ma; Paces and Miller, 1993) and Plešovice $(337.13 \pm 0.37 \text{ Ma}; \text{Sláma et al., } 2008)$ were analyzed as unknowns for 251

252	quality check. Data were acquired as batches of 6 unknowns bracketed by 4
253	measurements of 91500. Zircon U-Th-Pb data were processed using Glitter software
254	(Van Achterbergh et al., 2001) and age calculations were done using a present-day
255	238 U/ 235 U of 137.818 (Hiess et al., 2012) and decay constants of 1.55125.10 ⁻¹⁰ ,
256	9.8485.10 ⁻¹⁰ , and 4.9475.10 ⁻¹¹ yr ⁻¹ for 238 U, 235 U and 232 Th, respectively (Jaffey et al.,
257	1971; LeRoux and Glendenin, 1963). Concordia plots were built using the Isoplot
258	software (Ludwig, 2008). Finally, semi-quantitative concentrations of U, Th, and Pb
259	were determined by external normalization to 91500 as an additional means for
260	interpreting U-Pb ages, notably to detect correlation between ages and U concentrations
261	potentially arising from alteration facilitated by radiation-damage accumulation.

We obtained a Concordia age of 1097.5 \pm 4.5 Ma (2SE; n = 7; MSWD = 0.52) and weighted average ²⁰⁶Pb/²³⁸U and ²⁰⁷Pb/²⁰⁶Pb ages of 1096.9 \pm 6.6 Ma (2SE; n = 7; MSWD = 0.24), and 1097.3 \pm 7.7 Ma (2SE; n = 7; MSWD = 0.24,) respectively, for AS3, and a Concordia age of 338.2 \pm 2.9 Ma (2SE; n = 7; MSWD = 0.11), as well as 338.1 \pm 2.9 Ma (2SE; n = 7; MSWD = 0.14) and 343 \pm 22 Ma (2SE; n = 7; MSWD = 0.06), respectively, for Plešovice, which agree well with consensus ages (Wiedenbeck et al., 1995; Sláma et al., 2008). All results are available in Tables S5.

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270 **2.2.6.** Lu-Hf isotope measurements in Malawi zircon

Lu-Hf isotope measurements were done at LMV by LA-MC-ICP-MS using the same instruments as for Si isotopes. Details regarding operating conditions can be found in Table 2. Measurements were done next to traces of U-Pb and Si isotope analyses. Zircons were analyzed as batches of 10 unknowns (4 standards and 6 samples) bracketed by REE-absent MUN zircon standard (MUN-0; Fisher et al., 2011). Standards analyzed as unknowns (91500, MudTank, and two MUN zircons with low 277 and high Yb/Hf ratios) were used to monitor data quality. Yb and Hf instrumental mass biases were determined using ¹⁷³Yb/¹⁷¹Yb normalized to the value of 1.129197 278 (Vervoort et al., 2004) and ¹⁷⁹Hf/¹⁷⁷Hf normalized to 0.7325 (Stevenson and Patchett 279 280 1990). Lu fractionation was assumed to follow that of Yb (Fisher et al., 2014). Values of 0.793045 for ¹⁷⁶Yb/¹⁷³Yb (Vervoort et al., 2004) and 0.02655 for ¹⁷⁶Lu/¹⁷⁵Lu (Fisher 281 et al., 2014) were used to correct isobaric interferences of ¹⁷⁶Yb and ¹⁷⁶Lu at mass 176. 282 Since standards showed a slight shift, though reproducible, in ¹⁷⁶Hf/¹⁷⁷Hf after mass 283 bias correction, we have used MUN-0 to determine an offset factor (i.e., difference 284 between consensus and measured value) and correct all LA-MC-ICP-MS Lu-Hf 285 analyses accordingly. 286

287 Malawi zircons were measured during two separate analytical sessions. The first session gave average 176 Hf/ 177 Hf measured in 91500 and MudTank of 0.282312 \pm 288 289 0.000014 (2SD; n = 3) and 0.282508 ± 0.000023 (2SD; n = 2), respectively, and the 290 second session gave 0.282315 ± 0.000057 (2SD; n = 4) for 91500 and $0.282511 \pm$ 0.000023 (2SD; n = 4) for MudTank. These values compare well with consensus values 291 292 of 0.282308 ± 0.000008 (Blichert-Toft, 2008) and 0.282507 ± 0.000006 (Woodhead and Hergt, 2005). The average ¹⁷⁶Hf/¹⁷⁷Hf measured during the first session in MUN_1-293 294 2b and MUN_4-2b were 0.282139 ± 0.000037 (2SD; n = 5), and 0.282131 ± 0.000034 (2SD; n = 5), respectively. The average ¹⁷⁶Hf/¹⁷⁷Hf measured during the second session 295 in MUN 1-2b and MUN 3-2b were 0.282124 ± 0.000029 (2SD; n = 4), and 0.282127 296 297 \pm 0.000034 (2SD; n = 4), respectively. Therefore, our results on MUN zircons are fully 298 consistent with the value of 0.282135 ± 0.000007 reported in Fisher et al. (2011) which shows that our Lu-Hf data are accurately corrected for isobaric interferences created by 299 ¹⁷⁶Yb and ¹⁷⁶Lu. We used the ¹⁷⁶Lu decay constant of 1.867.10⁻¹¹ yr⁻¹from Scherer et 300 al. (2001) and Söderlund et al. (2004) to calculate initial Hf isotopic compositions. 301

302 Uncertainties associated with the radiogenic-ingrowth correction were propagated 303 using the algorithms of Ickert (2013) and ε_{Hf} values were calculated using CHUR values 304 provided in Iizuka et al. (2015). Results can be found in Table S6.

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306 **3. Results**

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3.1. Si isotopes in zircon

Figure 1 presents all Si isotope data in the form of δ^{30} Si values for zircon crystals 308 analyzed in this study by both laser-ablation MC-ICP-MS (colored squares; N = 188) 309 310 and solution MC-ICP-MS (black dots; N = 5). It can be seen that zircon Si isotope compositions exhibit significant variations at essentially three different scales; (1) at 311 312 the sample scale (within zircon population), (2) at the granite-type scale, and (3) 313 between granite types. In fact, zircon populations have either reproducible (e.g., MZ2, 314 40-03, Klodden, LyH; Fig. 1) or variable (e.g., Pikes Peak granite, AG09016, St-Julien, MARG; Fig. 1) δ^{30} Si values regardless of the type of granite considered. Similarly, not 315 all samples of a particular granite-type exhibit the same average zircon δ^{30} Si values. 316 Overall, zircon δ^{30} Si values are the highest in A-type granites (up to -0.13 ‰) and the 317 318 lowest in S-type granites (down to -1.19 %), with an overall decrease forming the 319 sequence A-type > TTG > I-type > S-type. However, there is too much overlap between 320 the data to give confidence to this pattern as is.

Zircons from A-type granites have δ^{30} Si values that vary from -0.67 to -0.11‰ (average = -0.39‰; Fig. 1; Tables 3 and S3), which translate into apparent fractionation factors between zircon and whole-rock (Δ^{30} Si_{WR-Zrc} = δ^{30} Si_{WR} - δ^{30} Si_{zrc}) of -0.04 to 0.52‰ (average = 0.24‰) when A-type bulk-rock δ^{30} Si of -0.15‰ is used (Savage et al., 2012). Zircons from TTGs have δ^{30} Si values that range from -0.83 to -0.29‰ (average = -0.50‰; Fig. 1; Tables 3 and S3), which translate into apparent fractionation

factors (Δ^{30} Si_{WR-Zrc}) of 0.29 to 0.83‰ (average = 0.50‰) when TTG bulk-rock δ^{30} Si 327 of 0.0% is used (Deng et al., 2019). Zircons from I-type granites have δ^{30} Si values that 328 vary from -0.91 to -0.27‰ (average = -0.52%; Fig. 1; Tables 3 and S3), which translate 329 into apparent fractionation factors (Δ^{30} Si_{WR-Zrc}) of 0.07 to 0.71‰ (average = 0.32‰) 330 when I-type bulk-rock δ^{30} Si of -0.2‰ is used (Savage et al., 2012). Zircons from S-331 type granites have δ^{30} Si values that range from -1.19 to -0.47‰ (average = -0.77‰; 332 Fig. 1; Tables 3 and S3), which translate into apparent fractionation factors (Δ^{30} Si_{WR-} 333 $_{\rm Zrc}$) of 0.17 to 0.89‰ (average = 0.46‰) when S-type bulk-rock δ^{30} Si of -0.3‰ is used 334 (Savage et al., 2012). Si isotope data obtained in this study and calculated apparent 335 fractionation factors are consistent with theoretical (Δ^{30} Si_{WR-Zrc} = 0.1 at 1300°C for an 336 337 andesitic melt and 0.45‰ at 600°C for a rhyolitic melt; Qin et al., 2016) and experimental results (Δ^{30} Si_{WR-Zrc} = 0.32 to 0.56‰ at 700°C; Trail et al., 2019), as well 338 as with available Si isotope data on minerals (zircon δ^{30} Si = -1.5 to 0.0%; Trail et al., 339 2018). 340

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3.2. Trace element concentrations

Results for trace elements are presented in Table S4 and Figures 2 and 3, which is 343 344 organized according to 4 groups (A-type, TTG, I-type, and S-type), and in Figures S3 and S4, which presents the same data but for each sample separately. Trace element 345 patterns in Figures 2 and S3 are normalized to chondrites (McDonough and Sun, 1995) 346 347 and data for zircons from each granite type are reported with AS3 data for comparison. In contrast, Figures 3, S2 and S4 are normalized to AS3. Analyzed zircons exhibit trace 348 element patterns that are consistent with that of AS3 (Fig. 2), though with two kinds of 349 350 variations. The first one is "coupled variation" which means that all trace element contents vary coherently with each other, and this results in a vertical shift of the entire 351

352 pattern without changing its shape (Fig. 2). This is precisely the case for rare earth elements (REE) from Gd to Lu. The second kind is "decoupled variation" which 353 354 corresponds to an increase in a chemical element content that is independent from the 355 pattern. This increase is, however, often correlated between several elements such that when one element content increases the content of other elements increase too but to 356 357 different extents. The best example corresponds to light REE (LREE) which show that 358 when La content increases both Ce, Pr, Nd, and Sm increase but to a lesser extent as 359 REE mass increases (Fig. 2). These two kinds of variations are visible in all groups but 360 not in all zircon crystals. Decoupled variation describes the behavior of elements that 361 are the most incompatible in zircon lattice (e.g., Al, Ca, Ga, Ba, LREE; Hoskin and 362 Schaltegger, 2003; Pidgeon et al., 2019), whereas coupled variation corresponds to the 363 behavior of least incompatible elements (e.g., Y, HREE, Hf, U). These two kinds of variations in trace element contents are shared by all granite types (Fig. 2). Zircon 364 365 chemistry is very similar between all granite types which makes it difficult to use in 366 granite-host identification as it was also acknowledged by Belousova et al. (2006). However, a few differences can be noted such as higher HREE content in S-type, an 367 368 overall shallower REE pattern in A-type, and higher Nb and Ta contents in A-type zircons that create shoulders in the patterns visible in Figure 2. Zircon patterns reveal 369 no or various extent of negative Eu and positive Ce anomalies, the latter being also 370 371 blurred by an increase in La contents in some cases. Due to the size of the laser-beam 372 and the scale of analyzed zircon crystals, variations in trace element contents and/or specific anomalies could not be associated with peculiar textural features, except in the 373 374 Malawi zircon in which domains of different origins could be identified more readily 375 (Figure 4). Figure 3 presents the same trace element data as in Figure 2 but normalized to AS3 which allows to more precisely observe the variation of chemical elements 376

around igneous values, represented by AS3, within each granite type as well as betweengranite types.

379 We determined Ti-in-zircon temperatures following the method of Ferry and 380 Watson, (2007). Activities of SiO_2 ($aSiO_2$) and TiO_2 ($aTiO_2$) play a fundamental role 381 in the validity of these temperature estimates which is why we used the software 382 Rhyolite-MELTS (Gualda et al., 2012) to determine the appropriate activities at the 383 zircon saturation temperature (T_{Zr}, Watson and Harrison, 1983; Miller et al., 2003; 384 Boehnke et al., 2013), according to suggestions of Schiller and Finger (2019). We used 385 T_{Zr} from both Watson and Harrison (1983) and Boehnke et al. (2013) calibrations as well as $T_{Zr/2}$ (i.e., T_{Zr} at half of the whole-rock Zr content; Schiller and Finger, 2019) 386 387 but this only resulted in temperature variations within a narrow range of about 20°C. 388 Similarly, changing the activity values within the main range of variation displayed in 389 Schiller and Finger (2019) does not drastically change the temperature estimates. 390 Furthermore, propagating all sources of uncertainty arising from the calibration curve 391 for Ti-in-zircon temperature (Ferry and Watson, 2007), that of zircon saturation 392 (Watson and Harrison, 1983; Boehnke et al., 2013), the input parameters and 393 calculations behind the Rhyolite-MELTS model (Gualda et al., 2012), and the 394 uncertainty on zircon Ti concentrations, we think that it seems unrealistic to expect a 395 Ti-in-zircon temperature more precise than $\sim 50^{\circ}$ C. Consequently, we have finally used 396 values of 0.95 for aSiO₂ and 0.45 aTiO₂, which are coherent conservative estimates of 397 the true activities (Schiller and Finger, 2019), to determine zircon crystallization temperatures and have considered these values with a 50°C uncertainty. The Ti-in-398 399 zircon temperatures obtained for zircon standards MudTank and AS3 comply well with 400 independent estimates from previous studies (Tables 3 and S4; Currie et al., 1992; Leu, 2016). Measured Ti-in-zircon temperatures gave 572-937°C for A-type granites (one 401

402 extreme at 1508°C), 716-974°C for TTGs (two extremes at 1107 and 1337°C), 758-970°C for I-type (two extremes at 1033 and 1312°C), and 776-1015°C for S-type 403 404 granites (Table S4). These temperatures usually vary within a range of $\leq 100^{\circ}$ C in each 405 sample, except for S-type granites in which determined temperatures vary by up to ~250°C. Ti-in-zircon temperatures, therefore, vary from one sample to the other but 406 407 without systematic difference between granite types. Temperature extremes mentioned 408 above always have Ti concentrations that correlate with Al, and/or K, and/or Ca, which 409 reach high values (e.g., 500-3000 ppm Al) that are inconsistent with elemental 410 substitution during magmatic growth (Hoskin and Schaltegger, 2003; Bouvier et al., 2012). Some Ti-in-zircon temperatures are consistent with zircon-saturation 411 412 temperatures (T_{zr}) whereas other disagree by more than 100°C (Table 3).

413

414 **3.3. Malawi zircon**

415 A back-scattered electron map of the megacrystic zircon (length = 1.5 cm) from Malawi 416 reveals a very well-defined oscillatory zoning from core to rim (Figure 4), nevertheless, 417 with visible concentric and radial fractures as well as metamorphic domains (dark grey zones) and biotite inclusions (black rounded areas). Various types of domains, labeled 418 MZ1-MZ7, have been analyzed to track possible δ^{30} Si variations and link them to 419 420 independent parameters such as crystallization (magmatic vs metamorphic) and 421 alteration conditions, chemical composition, age, Hf isotope signature, and 422 crystallization temperature. Traverses for Si isotope measurements by LA-MC-ICP-423 MS were done parallel to the magmatic zoning. Magmatic and metamorphic domains 424 returned identical U-Pb dates (Table S5) which gave a Concordia age of 726.6 ± 2.3 425 Ma (2SE; MSWD = 0.92), identical to that of 730 Ma reported by Ashwal et al. (2007) for megacrystic zircons from nepheline-bearing gneisses from Malawi. Metamorphic 426

427 and magmatic domains display sharp differences in Th/U and U concentrations, which is also visible in 176 Lu/ 177 Hf ratios, but more importantly in 176 Hf/ 177 Hf and δ^{30} Si 428 although differences are subtler in these isotopic systems (Figure 4). The chemical 429 430 distinction between magmatic and metamorphic domains also translates into an overall downward shift of trace element patterns to lower concentrations in Figure 2, and in Ti-431 in-zircon temperatures change from 627-743°C for magmatic to 572-602°C for 432 metamorphic domains (Table S4). The megacrystic zircon gave present-day ¹⁷⁶Hf/¹⁷⁷Hf 433 from 0.282464 ± 0.000018 (2SE) to 0.282560 ± 0.000024 (2SE; Table S6) regardless 434 of the domain analyzed, and ¹⁷⁶Lu/¹⁷⁷Hf from 0.0017 to 0.00004 with a sharp difference 435 between magmatic and metamorphic domains, the latter having the smallest ¹⁷⁶Lu/¹⁷⁷Hf 436 437 (Figure 4). Our new Lu-Hf isotope data are consistent with bulk-zircon solution data from Ashwal et al. (2007) that gave present-day 176 Hf/ 177 Hf of 0.282454 ± 0.000037 438 439 (2SE), 0.282466 ± 0.000057 (2SE), and 0.282475 ± 0.000057 (2SE) on three different zircon megacrysts for which the authors, however, did not measure ${}^{176}Lu/{}^{177}Hf. \delta^{30}Si$ 440 441 for magmatic domains gave an average value of $-0.43 \pm 0.15\%$ (2SD; average of 442 individual analyses) identical to the S-MC-ICP-MS value of $-0.46 \pm 0.10\%$ (2SD; Table 3), whereas metamorphic zones returned an average δ^{30} Si of -0.26 ± 0.19% 443 (2SD; average of individual analyses). Note that δ^{30} Si are very reproducible within each 444 445 magmatic domain and more variable in metamorphic domains (Figs 1 and 4). It should also be noted that slight variations in δ^{30} Si are visible between some of the magmatic 446 domains. In fact, MZ2 has lower δ^{30} Si than MZ1, MZ3, and MZ4 but similar to MZ5 447 which is patchy-looking in BSE but does not have metamorphic signatures as in MZ6 448 and MZ7 (Figs. 2 and 4). Interestingly, magmatic domains have very variable 449 176 Lu/ 177 Hf ratios that are, however, reproducible within each zone (Fig. 4). 450

452 **4. Discussion**

As presented in the result section, Si isotope signatures in zircon can be either reproducible or variable within a zircon population, hence, at the sample scale (Fig. 1). Before interpreting zircon δ^{30} Si in terms of fractionation behavior between crystals and their parental melts (primary processes), we need to, first, evaluate the effects of secondary processes such as metamorphic recrystallization and alteration/weathering, which can be assessed using zircon internal texture and chemical composition.

459

460 **4.1. Secondary causes of** δ^{30} Si variations at the sample scale

The megacrystic zircon from Malawi (MZ; Fig. 4) presents good evidence for 461 secondary modification of δ^{30} Si because metamorphic domains within this crystal 462 display a distinct δ^{30} Si value (-0.26‰) compared to magmatic domains (-0.46‰). 463 464 Metamorphic parts of MZ have typical characteristics of metamorphic zircons because 465 they appear dark and homogeneous in BSE images (Fig. 4; Corfu et al., 2003), have 466 very low Th/U (e.g., Vavra et al., 1999) and low Lu/Hf ratios, together with low trace element contents (Hoskin and Schaltegger, 2003). In addition, their estimated Ti-in-467 zircon temperature (~580°C) is lower than that of magmatic domains by about 100°C, 468 469 and in itself low for a magmatic zircon. Given that the degree of fractionation between 470 zircon and its parental melt/fluid increases as temperature decreases, the fact that metamorphic domains have higher δ^{30} Si than magmatic ones is good evidence for the 471 formation of metamorphic zircon domains from high- δ^{30} Si medium, possibly crustal 472 473 fluids (e.g., Kleine et al., 2018).

474 Metamorphism may also be responsible for the low δ^{30} Si of AG09-016 (TTG) 475 zircons (Fig. 5A) which exhibit a well-defined positive correlation between Th/U ratios 476 and δ^{30} Si values. Zircons from this 3947 Ma sample are known for having been affected 477 by metamorphism (e.g., Guitreau et al., 2018) but the ones analyzed for Si isotope were 478 selected to be the most pristine according to CL and BSE images, as well as U-Th-Pb 479 ages. This illustrates that metamorphism cannot always be straightforwardly detected 480 in zircon and each isotope system has its own sensitivity to metamorphism. When 481 considering the AG09-016 zircon that is least affected by metamorphism (i.e., with highest Th/U), its δ^{30} Si is very consistent with that of other TTG samples that have 482 reproducible zircon δ^{30} Si (Figs 1 and 5A). Metamorphism appears to have opposite 483 effects on δ^{30} Si values between MZ and AG09-016 zircons, which is likely controlled 484 485 by local factors (e.g., nature of fluids, temperature, water-rock ratio in crustal fluids) as well as what actually happens during metamorphism (i.e., recrystallization or enhanced 486 487 diffusion). The influence of metamorphism should therefore be evaluated on case-by-488 case basis.

489 Diffusion is a process known to possibly influence crystal elemental and isotope chemistry (e.g., Watson, 1996; Albarède, 2003; Cherniak and Watson, 2003) and we 490 491 can, therefore, wonder if Si diffusion can play a role in the observed variability in zircon 492 δ^{30} Si values (Fig. 1). Cherniak (2008) conducted diffusion experiments in zircon and 493 concluded that Si diffuses in zircon lattice at rates similar to those for Ti, which are 494 faster than those for U, Th, and Hf that are known to be essentially immobile in pristine 495 zircon (e.g., Cherniak and Watson, 2003), but slower than those for O. For example, it 496 would take 1.7 Ma for a silicon atom to diffuse over a radius of 1 μ m at 650°C. The 497 same distance would be reached after 12 Ga at 500°C. As a consequence, diffusion 498 cannot account for Si isotope variations in studied zircon crystals.

Despite zircon being very resistant to most geological processes, radioactive element decay (e.g., U and Th) can significantly damage its lattice over time which can lead to metamictization of growth bands or domains (e.g., Holland and Gottfried, 1955;

502 Murakami et al., 1991) and formation of cracks (Lee and Tromp, 1995), altogether making zircon sensitive to secondary alteration (e.g., Corfu et al., 2003; Guitreau and 503 Flahaut, 2019; Pidgeon et al., 2019). Since alteration and weathering can influence 504 505 zircon chemical and isotopic compositions (e.g., Bouvier et al., 2012; Pidgeon et al., 506 2017, 2019; Guitreau and Flahaut, 2019) and that it has an effect on rocks and minerals 507 (e.g., Ziegler et al., 2005; Opfergelt et al., 2012), we used trace element contents (i.e. non-formula elements) to assess zircon pristineness. The measured elevated contents of 508 509 non-formula elements (e.g., Ca, K, Al) and their possible co-variation in some zircon crystals (Figs. 2 and S3; Table S4) testify that alteration/weathering occurred in some 510 511 of the studied crystals (Pidgeon et al., 2019). However, most zircon populations displayed no relationship whatsoever between δ^{30} Si values and the concentration of 512 513 alteration/weathering proxies. The 1.94 Ga (I-type) sample Knaften is one of the rare examples in which alteration/weathering seems to have potentially influenced the δ^{30} Si 514 515 value of one zircon which is associated with a high Ca concentration (Fig. 5B). It is 516 also evident in this sample that alteration/weathering is responsible for the high Ti concentration of the same zircon as illustrated by the positive correlation between Ca 517 518 and Ti (Fig. 5C). This, in turn, means that the calculated Ti-in-zircon temperature for 519 this crystal is over-estimated and no longer represents that of igneous crystallization. 520 This testifies to the spurious nature of the extreme Ti-in-zircon temperatures mentioned 521 in the result section as in all these cases Ti content correlates with that of light elements 522 such as Al, Ca, K, (Table S4; Figs 2 and S3) which is diagnostic of external material 523 input into zircon (e.g., Pidgeon et al., 2019).

As a consequence, alteration/weathering can affect zircon chemistry but does not always lead to modifications of zircon Si isotope signatures. However, this assertion should be regarded with some caution because Si isotopes were measured using

527 traverses, that covered a significant portion of each crystal, whereas trace element 528 measurements were done using spots that were located unto the marks left by Si isotope 529 analyses. Therefore, measurements were performed sequentially and on distinct 530 material although spatially related. This is the major limitation of our approach and it is why the absence of significant influence of alteration/weathering on Si isotopes 531 532 should not be considered as a firm conclusion. Nevertheless, our data still allow a fairly good understanding of the relationship between δ^{30} Si and trace elements since most of 533 534 the analyzed zircons were pre-selected based on pristine-looking internal textures (fine 535 oscillatory zoning) and existing chemical and/or isotopic data. It is, consequently, 536 possible that a crystal was locally altered, which was detected by trace elements, 537 although overall essentially pristine thereby giving non-altered Si isotope signatures 538 that we can further discuss in an igneous context.

539

540 **4.2. Primary causes of** δ^{30} Si variations at the sample scale

541 Silicon isotope fractionation between zircon and its parental melt mostly depends on the temperature of zircon crystallization ($\Delta^{30}Si_{WR-Zrc} \propto 10^6/T^2$), and the degree of 542 543 polymerization of SiO₄ in the melt (e.g., Grant, 1954; Poitrasson and Zambardi, 2015), which can be approached by the NBO/T parameter (Mysen et al., 1985) and SiO₂ 544 545 content (e.g., Qin et al., 2016; Poitrasson, 2017; Trail et al., 2019). Ti-in-zircon 546 temperature (Ferry and Watson, 2007) offers a good proxy for the crystallization 547 temperature of zircon crystals and the vast majority of our samples have Ti-in-zircon 548 temperatures that vary within ~100°C. According to experimental calibrations from Trail et al. (2019), δ^{30} Si variation only related to 100°C drop would go from 0.07‰ 549 between 900° and 800°C to 0.14‰ between 700° and 600°C. This is in the case of 550 zircon-quartz equilibration (SiO₂ concentration of ~100 wt%) for which the A value of 551

the fractionation equation (i.e., Δ^{30} Si_{Qz-Zrc} = A x 10⁶/T²) is 0.53 ± 014 (Trail et al., 2019). 552 These fractionation values of 0.07 and 0.14‰ should, therefore, be regarded as 553 554 maximum since the degree of fractionation between zircon and melt would be smaller 555 for lower SiO₂ contents (Qin et al., 2016; Trail et al., 2019). These variations are smaller 556 or equivalent to the analytical uncertainty of our analyses and would, hence, be undetected. Consequently, samples that display reproducible zircon δ^{30} Si associated 557 with narrow Ti-in-zircon temperatures very likely formed from magmas with 558 559 homogeneous SiO₂ contents (e.g., SKJ, 40-03, Klodden; Figs. 1 and 6) and 560 homogeneous Si isotope compositions.

In contrast, samples with variable zircon δ^{30} Si can be due to crystallization over 561 562 large temperature windows (>100°C) and/or from melts with variable SiO₂ contents 563 either locally (e.g., mingling effects) or temporal (e.g., fractional crystallization 564 effects). Figure 5D presents the only relationship between Ti-in-zircon temperatures and δ^{30} Si values visible in our dataset. It likely comes from the fact that this sample, 565 566 MARG - a cordierite-bearing (S-type) granite - is the only sample that has a Ti-inzircon temperature range of ~250°C (1015-775°C; Table S4). The relationship between 567 δ^{30} Si values and temperature complies with theory and experiments but the degree of 568 fractionation (0.72‰) is too large because a maximum δ^{30} Si variation of 0.16‰ should 569 570 be expected if only temperature controlled the fractionation (Figs. 5D and 6D; Trail et 571 al., 2019). In order to account for the excess fractionation, we can invoke a change in 572 SiO₂ content of the parental melt but it does not entirely solve the problem (Figure 6D) because the degree of fractionation of these zircons, expressed as Δ^{30} Si_{WR-Zrc}, exceed 573 plausible values when a δ^{30} Si of -0.3% is assumed for S-type granites (Savage et al., 574 2011). A way to account for this strong evolution of MARG zircon δ^{30} Si is a source 575 576 control because this granite is a crust-reworking product that involved clastic sediments

577 (e.g., Couturié, 1977), and low-temperature alteration of crustal rocks to form clay minerals result in the development of low δ^{30} Si signatures down to -2‰ (e.g., Ziegler 578 et al., 2005; Opfergelt et al., 2012), opposite from what is seen with O isotopes (e.g., 579 Valley, 2003; Bindeman, 2008). An additional consideration to account for the very 580 large spread in δ^{30} Si and Ti-in-zircon temperature is to consider that some of the 581 582 analyzed zircons, or some parts of these zircons, could be inherited since this is a common feature of crustally-derived granitoids (e.g., Villaros et al., 2012; Couzinié et 583 584 al., 2021). This would, in turn, account for the very high temperature of 1015° C, which is hard to reconcile with the S-type nature of the Margeride granite. 585

The remaining samples with variable zircon δ^{30} Si have narrow Ti-in-zircon 586 587 temperatures (e.g., Pikes Peak, Kristineberg; Figs. 1 and 6) which, therefore, cannot 588 account for this within-zircon-population variability. The most likely explanation is 589 local heterogeneities in SiO₂ concentrations and/or temperatures coming from magma mingling and/or mixing, or simply fractional crystallization. At 800°C, the difference 590 between the δ^{30} Si of zircons formed from mafic (SiO₂ = 50 wt%) and felsic magmas 591 $(SiO_2 = 70 \text{ wt\%})$ would be ~0.3‰ (Figure 6; Qin et al., 2016). These melts would most 592 likely have distinct temperatures which would further increase the variability in δ^{30} Si 593 594 values. One would have wished that zircon chemistry could help further support this 595 hypothesis. However, trace element contents in zircon do not always significantly vary 596 between crystals derived from mafic and felsic melts (e.g., Hoskin and Schaltegger, 597 2003), despite strong elemental variations between these two magma types. This, 598 unfortunately, makes zircon a poor indicator of magma origin, except perhaps for A-599 type (Fig. S5), and account for the general absence of correlation between elemental concentrations and δ^{30} Si values. Interestingly, magma mingling has been reported for 600 some of the samples exhibiting variations in zircon δ^{30} Si values. This is the case for 601

Kristineberg that belongs to the well-studied Jörn granitoid group for which Wilson et
al. (1987) described multiple magma sources and interactions. Similarly, St-Julien-laVêtre granite is also known for obvious magma mingling between mafic and felsic endmembers (Barbarin, 1988a).

A source control can also be invoked to account for some zircon $\delta^{30}Si$ 606 607 variations, for example in S-type MARG (already discussed above), I-type Knaften and 608 Kristineberg (Figure 6). In fact, Wasström (1993) described Knaften granitoids as 609 subvolcanic plutons that intruded into mafic rocks and a supracrustal sequence 610 containing greywacke, turbidite, and graphite-bearing argillites. If these supracrustals interacted with Knaften magmas, it would have lowered melt δ^{30} Si which could account 611 for the sharp increase in Knaften $\Delta^{30}Si_{WR-Zrc}$ below 850°C that fall slightly above 612 maximum fractionation (SiO₂ \sim 100 wt%; Fig. 6C) as calculated using a I-type bulk-613 614 rock δ^{30} Si value of -0.2‰. The typical A-type Pikes Peak granite involved both mantle-615 derived and crustally-derived, though not sedimentary, components (Smith et al., 1999; 616 Guitreau et al., 2016) which likely resulted in magmas with variable SiO₂ contents, though with similar δ^{30} Si values. Silicon isotope composition of zircon, therefore, 617 appears to be sensitive to complex magma evolution and track source changes and 618 619 magma interactions.

Returning to the Malawi zircon example, magmatic domains MZ1, MZ3 and MZ4 have identical δ^{30} Si values, whereas MZ2, also magmatic, has a lower δ^{30} Si value (Figs. 1 and 4) despite MZ2 being located between growth bands MZ4 (innermost) and MZ3-MZ1 (outermost and being from the same growth bands but located in opposite parts of the zircon; Fig. 4). Our results also show that MZ2 crystallized at higher temperature than MZ3 and from a magma with apparently higher SiO₂ content (Table 3; Fig. 6). In addition, the central domain of the zircon (MZ5) that was analyzed but

627 interpreted as intermediate between magmatic and metamorphic because of patchylooking internal texture (Figs. 1 and 4), has similar δ^{30} Si value as MZ2, and most of the 628 trace element patterns, alongside Ti-in-zircon temperatures, are magmatic-looking (Fig. 629 630 2; Table 3). Therefore, MZ5 can be interpreted as the oldest magmatic growth zone of MZ and the coupled δ^{30} Si-temperature evolution from core (MZ5) to borders (MZ1-631 632 MZ3) reveal cycles of temperature increase and decrease associated with fluctuations in SiO₂ content over the course of zircon crystallization, possibly accounting for the 633 634 large size of this crystal. These changes in magma composition, and/or crystallization temperature are consistent with the fact that ¹⁷⁶Lu/¹⁷⁷Hf ratio vary significantly between 635 growth zones but this could also reveal equilibration with co-precipitating phases that 636 637 changed during zircon growth.

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4.3. Zircon δ^{30} Si variations within and between granite types

640 So far, our results combined with theoretical and experimental studies (Qin et al., 2016; 641 Trail et al., 2019) allowed us to demonstrate that secondary processes, such as metamorphism and alteration/weathering, can modify zircon δ^{30} Si values, which could 642 lead to an erroneous interpretation of the zircon parental melt composition, and that 643 644 zircon-melt isotope fractionation is sensitive to crystallization temperature and magma silica content. Therefore, variations of zircon δ^{30} Si between samples of the same granite 645 type as visible in Figure 1 can be accounted for by differences in zircon crystallization 646 647 temperatures, whole-rock SiO₂ contents, and magma heterogeneities (i.e., Si elemental and isotopic compositions) linked to fractional crystallization and/or magma mingling 648 and/or source controls as illustrated in Figure 6 and Table 3. This is for example the 649 650 case between SKJ and Malawi magmatic domains because the former is higher temperature than the latter, though they have almost identical parental magma SiO_2 651

652 content (Fig. 6A; Table 3). Inversely, Pikes Peak granite has higher whole-rock SiO₂ content than SKJ, though similarly high temperatures (Fig. 6A). Klodden and 653 654 Kristineberg belong to the same type of pluton (Jörn-type; Billström and Weihed, 1996) and have overlapping zircon δ^{30} Si values but they extend to lower values in 655 656 Kristineberg (Figs 1 and 6C). This is well explained by the fact that Klodden has lower 657 SiO₂ content than Kristineberg, in which zircon likely crystallized from evolving melts, hence, with changing SiO₂ content (Fig. 6C). Vargfors shows lower zircon 658 659 crystallization temperatures and higher SiO₂ content than Klodden, which accounts for the differences in zircon δ^{30} Si (Table 3; Figs 1 and 6C). In contrast, differences visible 660 between TTG zircon δ^{30} Si values are mostly due to metamorphic imprints because 661 662 Steynsdorp (Stp) zircons exhibit similar relationship between δ^{30} Si values and Th/U ratios as AG09-016 zircons (Fig. 5A; Tables S3 and S4). Moreover, Napier zircons 663 664 (78285007 and 78285013) are known for having been severely affected by metamorphism (e.g., Guitreau et al., 2019). It should also be noted that TTG samples 665 666 studied in this contribution are the same, or equivalents from the same outcrops/plutons, as those analyzed for bulk-rock Si isotope compositions in Deng et al., (2019), except 667 Napier samples 78285007 and 78285013, and they all gave a very consistent average 668 bulk-rock δ^{30} Si value of -0.01 ± 0.02‰ (2 SD). 669

670 When all factors controlling zircon δ^{30} Si values are considered, one can 671 understand why data from various granite types overlap in Figure 1. It is in fact because 672 the range of zircon δ^{30} Si arising from fractionation related to melt composition (SiO₂ 673 and δ^{30} Si) and temperature variation could be larger than the difference between bulk-674 rock δ^{30} Si values. This is also close to the analytical precision and/or reproducibility of 675 Si isotope measurements by LA-MC-ICP-MS and S-MC-ICP-MS (Table 3) which 676 tends to increase natural variations. Yet, once all sources of variations are controlled and secondary signatures corrected for, zircon δ^{30} Si values comply relatively well with bulk-rock δ^{30} Si pattern, as illustrated in Figure 7. This figure also reveals that the current analytical precision achieved by LA-MC-ICP-MS does not prevent this technique to be used but is a limiting factor that should be improved in the future.

681 A very good illustration of the misleading interpretation possibly arising from 682 interpreting Si isotopes in zircon at face value is given by the two non-coeval Acasta samples AG09009 and AG09008g. The former and the latter belong to TTG (bulk-rock 683 δ^{30} Si of 0.0%; André et al., 2019; Deng et al., 2019) and I-type (bulk-rock δ^{30} Si of -684 0.2%; Savage et al., 2012; Poitrasson and Zambardi, 2015), respectively, which is 685 visible from both their bulk-rock chemical composition (Mojzsis et al., 2014) and their 686 687 bulk-rock δ^{30} Si value (Deng et al., 2019). Despite these differences, they exhibit identical (and reproducible) zircon δ^{30} Si values of approximately -0.4% (Fig. 1; Table 688 689 $\frac{3}{2}$). This is well accounted for by the fact that AG09-009 has higher SiO₂ content than AG09-008g and its zircons have lower Ti-in-zircon (crystallization) temperatures than 690 691 those from AG09-008g. These two factors together result in an increase of the isotope fractionation between zircons and their parental melt (Δ^{30} Si_{WR-Zrc}) in AG09-009 692 693 compared to AG09-008g. Consequently, without knowledge of SiO₂ content and 694 crystallization temperatures, it is virtually impossible to discriminate these Acasta 695 samples from each other. In order for Si isotopes in zircon to be used effectively in out-696 of-context zircons (i.e., detrital crystals), a proxy for SiO₂ content or of the nature of 697 the zircon parental melt should be used, as also pointed out in Qin et al. (2016) and 698 Trail et al. (2019).

We took advantage of our dataset to calibrate the relationship between SiO₂ content and the A value, which was done indirectly by comparison with O isotopes in Qin et al. (2016) and using data from Trail et al. (2018) in Trail et al. (2019). Figure 8

presents the relationship between A value ($\Delta^{30}Si_{WR-Zrc}/10^6/T^2$) as a function of melt SiO₂ content that can be tentatively used to build more accurate figures such as those in Figure 6. We obtain the following relationship:

705
$$\Delta^{30}Si_{WR-Zrc} = \frac{10^6}{T^2} \times 0.00545x$$

706 With x being the whole-rock SiO_2 content in wt%, supposedly equivalent to that of the 707 parental melt, and T the temperature in Kelvin (Fig. 8). Zircon crystallization 708 temperature can be either obtained from Ti-in-zircon thermometer (Ferry and Watson, 709 2007) or possibly from zircon-saturation (Watson and Harrison, 1983; Miller et al., 710 2003). However, we observed in this study that Ti-in-zircon temperatures are quite 711 consistent with zircon-saturation temperatures (Tzr; Table 3) in some cases (e.g., TTGs, 712 MZ), uncertainty considered, but are quite different in others (e.g., SKJ, Klodden, 713 Kristineberg). This mismatch is strongest in studied I-type granites, which illustrates 714 the subjective character of T_{zr} that most likely come from the fact that granites do not 715 often represent actual magmatic liquids. Therefore, Ti-in-zircon temperatures should 716 be preferred because, contrary to T_{zr}, they can actually track zircon parental magma evolution. 717

718

719 **5.** Conclusions

This study focuses on the Si isotope composition of zircons from various granite types. We show that zircon δ^{30} Si can vary significantly at the sample scale (i.e., within a zircon population) but also between granites of the same type and between granite types. These variations are due to differences in zircon crystallization temperatures and parental magma SiO₂ contents, which makes zircon δ^{30} Si a good proxy for magma evolution. However, secondary processes, such as metamorphism and alteration/weathering, can affect zircon δ^{30} Si values but they can be readily identified on the basis of zircon internal textures, as revealed by BSE/CL images, and trace-element geochemistry.
Therefore, Si isotopes in zircon have promising applications to deciphering granitic
magma evolution as well as granite-type discrimination that can be improved even
further by reducing analytical uncertainty which is currently the major limitation of this
technique.

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Stable isotope geochemistry of silicon in granitoid zircon

Table 1. Sample details

Table 2. LA-ICP-MS and LA-MC-ICP-MS operating conditions

Table 3. Average Si isotope signatures of zircons analyzed in this study

Table 1

Gr	anite type*	Origin	Name	Rock type	Age (Ma)	Reference					
A-type	PAG	Tambani (Malawi)	MZ	Syenite (nepheline-bearing)	730	Ashwal et al., 2007					
A-type	PAG	Skjoldungen (Greeland)	SKJ	Syenite	2690	Rosing et al., 1988; Nutman and Rosing, 1994					
A-type	PAG	Pikes Peak batholith (USA)	YPP-20	Syenogranite	1066	Smith et al., 1999; Guitreau et al., 2016					
A-type	PAG	Cripple Creek batholith (USA)	EMQ-5	Quartz-monzonite	1432	Guitreau et al., 2016					
TTG	~ATG/ACG	Sharyzhalgai uplift (Siberia)	40-03	Trondhjemite	3400	Guitreau et al., 2012					
TTG	~ATG/ACG	Acasta Gneiss Complex (Canada)	AG09009	Granodiorite	3600	Guitreau et al., 2012					
TTG	~ATG/ACG	Karelian craton (Finlande)	H395	Trondhjemite	2800	Guitreau et al., 2012					
TTG	~ATG/ACG	Barberton (South Africa)	NLS7.2	Trondhjemite	3245	Guitreau et al., 2012					
TTG	~ATG/ACG	Sharyzalgai uplift (Siberia)	58-03	Tonalite	3392	Guitreau et al., 2012					
TTG	~ATG/ACG	Barberton (South Africa)	Steynsdorp	Trondhjemite	3518	Guitreau et al., 2012					
TTG	~ATG/ACG	Napier Complex (Antarctica)	78285007a	Tonalite	3794	Guitreau et al., 2012, 2019					
TTG	~ATG/ACG	Napier Complex (Antarctica)	78285013	Granite	3857	Guitreau et al., 2012, 2019					
TTG	~ATG/ACG	Acasta Gneiss Complex (Canada)	AG09016	Tonalite	3947	Guitreau et al., 2012					
I-type	ATG	Acasta Gneiss Complex (Canada)	AG09008g	Tonalite	3975	Guitreau et al., 2012, 2014a, 2018; ; Mojzsis et al., 2014					
I-type	ACG	Skellefte (Sweden)	Klodden	Granodiorite	1902	Guitreau et al., 2014b					
I-type	KCG	Skellefte (Sweden)	Barsele	Dacite	1959	Guitreau et al., 2014b					
I-type	ACG	Skellefte (Sweden)	Melestjärn	Quartz-felsdpar porphyritic flow	1889	Guitreau et al., 2014b					
I-type	ACG	Skellefte (Sweden)	Kristineberg	Tonalite	1907	Guitreau et al., 2014b					
I-type	KCG	St-Julien-la-vêtre, Forez (France)	St-Julien	Monzogranite	340	Didier et al., 1989					
I-type	ACG	Skellefte (Sweden)	Vargfors	Ignimbite associated to granitoids	1875	Guitreau et al., 2014b					
I-type	ATG	Acasta Gneiss Complex (Canada)	AG09017	Tonalite	3956	Guitreau et al., 2012, 2018; Mojzsis et al., 2014					
I-type	ACG	Skellefte (Sweden)	Bjurvattnett	Sub-volcanic porphyritic rhyodacite	1884	Guitreau et al., 2014b					
I-type	KCG	Skellefte (Sweden)	Knaften	Granodiorite	1940	Guitreau et al., 2014b					
S-type	CPG	Margeride (France)	MARG	Monzogranite	323	Couturié et al., 1979					
S-type	MPG	Hermitage, Forez (France)	LγH	Monzogranite	325	Didier et al., 1989					
S-type	CPG	Meymac (France)	Meymac	Monzogranite	295	Talbert and Duthou, 1983					

* I-type = Infracrustal/Igneous, S-type = Supracrustal, A-type = Alkaline/Anorogenic, TTG = Tonalite, Trondhjemite, Granodiorite, PAG = Peralkaline and Alkaline Granitoids, ATG = Arc Tholeiitic Granitoids, ACG = Amphibole-bearing Calc-alkaline Granitoids, KCG = K-rich Calc-alkaline Granitoids, CPG = Cordierite-bearing Peraluminous Granitoids, MPG = Muscovite-bearing Peraluminous Granitoids)

Table 2

Analysis		Si isotopes				L	.u-Hf i	sotope	es			U-Pb geochronology	Trace element concentrations			
Laboratory	Мад	Magmas et Volcans				Ма	gmas	et Vold	ans			Magmas et Volcans	Magmas et Volcans			
Cup configuration	figuration L3 C H3 L4 L3 L2 L1 C H1 H2 H3		N.A.	N.A.												
Mass measured	28	29	30	17 1	17 3	17 4	17 5	17 6	17 7	17 8	17 9	204, 206, 207, 208, 232, 238	27-29, 39, 43, 44, 49, 51, 66, 71, 85, 88-91, 93, 96, 137, 139-141, 146, 147, 153, 157, 159, 163, 165, 166, 169, 172, 175, 178, 181, 206, 232, 238			
ICP-MS model	Therr	mo Neptu	ne <i>Plus</i>			The	rmo Ne	eptune	Plus			Thermo Element XR	Thermo Element XR			
Interface type	Jet	(high sens	sitivity)			Jet	(high :	sensiti	/ity)			Jet (high sensitivity)	Jet (high sensitivity)			
Cones		X-Cones	6				X-C	ones				X-Cones	X-Cones			
Resistors		10 ¹¹ Ohm	IS				10 ¹¹ (Ohms				Counting/Analog	Counting/Analog			
Resolution	Hig	h (M/∆M=	7800)	Low								Low	Low			
Forward Power		1200 W		1200 W								1300 W	1300 W			
Auxiliary gas (Ar)	().7-0.8 L/r	nin	0.72 L/min								~0.7 L/min	0.75 L/min			
Ar sample		~1 L/mir	ı	0.97 L/min								~1.0 L/min	1.01 L/min			
Additional gas (N ₂)		N.A.		4.0 mL/min								7.5 mL/min	6.5 mL/min			
Oxide ThO/Th		N.A.					N	.A.				~1%	~1%			
SQUID		On		On								On	On			
Data acquisition	Time resolved analysis			Time resolved analysis								Time resolved analysis	Time resolved analysis			
Scanning mode	Static multi-collection Static multi-collection										Peak-jumping	Mass scan				
Integration time per isotope		1s		1s								5 ms	5 ms			
Background counting		20-30 s		20-30 s								20-30 s	20-30 s			
time Sample measurement time									60 s	60 s						
Measurement type	Star	ndard brac	keting	Internal and external normalization								Standard bracketing Standard bracketing				
External standard	91500, KIM, AS3,					500, M	lud Tai	nk, MU	N zirco	ons		91500, AS3, Plešovice	NIST 610, NIST 612, 91500, MudTank AS3			
Laser model	Reson	etics Reso 50E	olution M-	Resonetics Resolution M-50E Resonetics Resolution 50E				Resonetics Resolution M- 50E	Resonetics Resolution M-50E							

Wavelength	193 nm	193 nm	193 nm	193 nm
Pulse duration	5 ns	5 ns	5 ns	5 ns
Energy	2-5 mJ	~3.2 mJ	~3.5 mJ	~3.5 mJ
Fluence	~3 J/cm2	~3.6 J/cm2	~2.5 J/cm2	~2.5 J/cm2
Frequency	6 Hz	6 Hz	4 Hz	2 Hz
Spot size	33 µm	40 µm	33 µm	33 µm
Carrier gas	Не	Не	He	Не
Carrier gas flow	0.65-0.75 L/min	0.7 L/min	0.65 L/min	0.75 L/min
Traverse rate	1.6 µm/s	N.A.	N.A.	N.A.

Table 3

			Relative to NBS 28]							
MC-ICP-MS method	Granite type	Name	δ29Si (‰)	2SD	δ30Si (‰)	2SD	n	∆ ³⁰ Si _{WR-zrc} (‰)	WR SiO ₂	TZr	TZr*	Zircon T (°C)	10 ⁶ /T ²	А
Laser ablation	A-type	MZ1	-0.25	0.11	-0.38	0.13	6	0.23	58.10	560	626	-	-	-
Laser ablation	A-type	MZ2	-0.24	0.10	-0.54	0.05	6	0.39	58.10	560	626	699	1.06	0.37
Laser ablation	A-type	MZ3	-0.21	0.11	-0.41	0.08	6	0.26	58.10	560	626	649	1.18	0.22
Laser ablation	A-type	MZ4	-0.24	0.06	-0.40	0.12	6	0.25	58.10	560	626	-	-	-
Laser ablation	A-type	MZ5	-0.25	0.07	-0.50	0.05	6	0.35	58.10	560	626	657	1.16	0.30
Laser ablation	A-type	MZ6	-0.17	0.10	-0.29	0.22	6	0.14	58.10	-	-	-	-	-
Laser ablation	A-type	MZ7	-0.14	0.08	-0.24	0.17	6	0.09	58.10	-	-	580	1.37	0.06
Laser ablation	A-type	SKJ	-0.16	0.14	-0.22	0.11	6	0.07	57.73	630	695	828	0.82	0.08
Laser ablation	A-type	Pikes Peak	-0.22	0.11	-0.43	0.27	6	0.28	68.07	874	895	911	0.71	0.39
Laser ablation	A-type	EMQ-5	-0.31	0.05	-0.55	0.18	6	0.40	60.98	834	862	-	-	-
Laser ablation	TTG	40-03	-0.18	0.09	-0.37	0.10	5	0.37	71.31	716	756	758	0.94	0.39
Laser ablation	TTG	AG09009	-0.23	0.09	-0.40	0.13	5	0.40	71.62	828	852	799	0.87	0.46
Laser ablation	TTG	H395	-0.27	0.12	-0.43	0.16	5	0.43	68.80	810	836	-	-	-
Laser ablation	TTG	NLS7.2-08	-0.17	0.12	-0.39	0.18	5	0.39	66.68	738	778	-	-	-
Laser ablation	TTG	58-03	-0.25	0.16	-0.46	0.15	5	0.46	69.43	764	802	-	-	-
Laser ablation	TTG	Stp	-0.29	0.10	-0.57	0.14	5	0.57	73.91	771	805	854	0.79	0.72
Laser ablation	TTG	78285007a	-0.36	0.11	-0.60	0.20	5	0.60	66.90	640	695	-	-	-
Laser ablation	TTG	78285007na	-0.28	0.19	-0.55	0.17	6	0.55	66.90	640	695	-	-	-
Laser ablation	TTG	78285013	-0.37	0.10	-0.57	0.17	7	0.57	71.88	602	661	1157	0.49	1.18
Laser ablation	TTG	AG09016	-0.27	0.14	-0.59	0.27	7	0.59	66.23	918	925	933	0.69	0.86
Laser ablation	I-type	AG09008g_8	-0.22	0.12	-0.42	0.10	7	0.22	56.71	790	835	-	-	-
Laser ablation	I-type	Klodden	-0.23	0.07	-0.40	0.10	6	0.20	57.75	641	692	888	0.74	0.27
Laser ablation	I-type	Barsele	-0.17	0.12	-0.35	0.25	5	0.15	68.10	824	838	-	-	-
Laser ablation	I-type	Melestjärn	-0.28	0.16	-0.46	0.42	5	0.26	75.03	767	798	-	-	-
Laser ablation	I-type	Kristineberg	-0.29	0.12	-0.49	0.26	5	0.29	62.10	725	769	933	0.69	0.42

Laser ablation	I-type	St-Julien	-0.33	0.25	-0.62	0.30	5		0.42	64.80	-	-	-	-		-
Laser ablation	I-type	Vargfors	-0.28	0.10	-0.53	0.17	6		0.33	71.97	737	776	823		0.83	0.40
Laser ablation	I-type	AG09017	-0.31	0.14	-0.61	0.21	7		0.41	60.16	774	808	-	-		-
Laser ablation	I-type	Bjurvattnett	-0.35	0.11	-0.65	0.18	6		0.45	69.35	770	798	-	-		-
Laser ablation	I-type	Knaften	-0.31	0.23	-0.69	0.33	6		0.49	74.60	710	753	907		0.72	0.68
Laser ablation	S-type	MARG	-0.45	0.29	-0.80	0.59	4		0.50	68.23	-	-	906		0.72	0.70
Laser ablation	S-type	LYH	-0.38	0.12	-0.74	0.11	5		0.44	74.20	-	-	-	-		-
Laser ablation	S-type	Meymac	-0.41	0.08	-0.75	0.18	6		0.45	70.38	-	-	831		0.82	0.55
Solution	TTG	40-03-Zrc	-0.21	0.09	-0.42	0.13	4		0.42	71.31	716	756	758		0.94	0.45
Solution	TTG	H395-Zrc	-0.19	0.03	-0.41	0.07	4		0.41	68.80	810	836	-	-		-
Solution	I-type	Kristineberg-Zrc	-0.21	0.07	-0.42	0.10	6		0.22	62.10	725	769	933		0.69	0.32
Solution	A-type	EMQ-5-Zrc	-0.22	0.03	-0.40	0.08	3		0.25	60.98	834	862	-	-		-
Solution	A-type	MZ-2	-0.20	0.02	-0.46	0.10	4		0.31	58.10	560	626	699		1.06	0.29
Laser ablation	Standard	AS3	-0.25	0.14	-0.47	0.20	16		0.18	50.00	-	-	970		0.65	0.27
Laser ablation	Standard	KIM	-0.21	0.13	-0.41	0.19	12		0.12	35.00	-	-	678		1.11	0.11
Laser ablation	Standard	MudTank	-0.19	0.11	-0.38	0.23	5		0.09	15.00	-	-	789		0.89	0.10
Laser ablation	Standard	Plešovice	-0.23	0.08	-0.44	0.21	5		0.24	61.00	-	-	-	-		-
Laser ablation	Standard	MUN-0	-0.84	0.14	-1.61	0.21		-		-	-	-	-	-		-
Laser ablation	Standard	MUN-1	-1.04	0.15	-2.02	0.27	16	-			-	-	-	-		-

 Δ^{30} Si_{WR-zz} is calculated using δ^{30} Si_{WR} of -0.15‰ for A-type, 0.0‰ for TTG, -0.2‰ for I-type, -0.3‰ for S-type, and -0.29‰ for AS3, KIM and MudTank.

Tzr calculated using the calibration of Boehnke et al. (2013) whereas Tzr* was calculated using the original calibration of Watson and Harrison (1983)

Zircon T°C is the average Ti-in-zircon temperature determined for each sample

Temperature is expressed in Kelvin in 10⁶/T²

A is the constant in the fractionation law: $\Delta^{30}Si_{WR-zrc} = A^*10^6/T^2$

Stable isotope geochemistry of silicon in granitoid zircon

Figures

Figure 1. Si isotope results for zircon crystals analyzed in this study by S-MC-ICP-MS (black spots) and LA-MC-ICP-MS (colored squares) in the form of δ^{30} Si values. Also indicated are the average δ^{30} Si for mantle zircon (-0.38 ± 0.04 ‰; 2SD; Trail et al., 2018) together with bulk-rock average δ^{30} Si values for major granite types (A-, I-, S- and TTGs; Savage et al., 2012; Poitrasson and Zambardi et al., 2015; Deng et al., 2019).

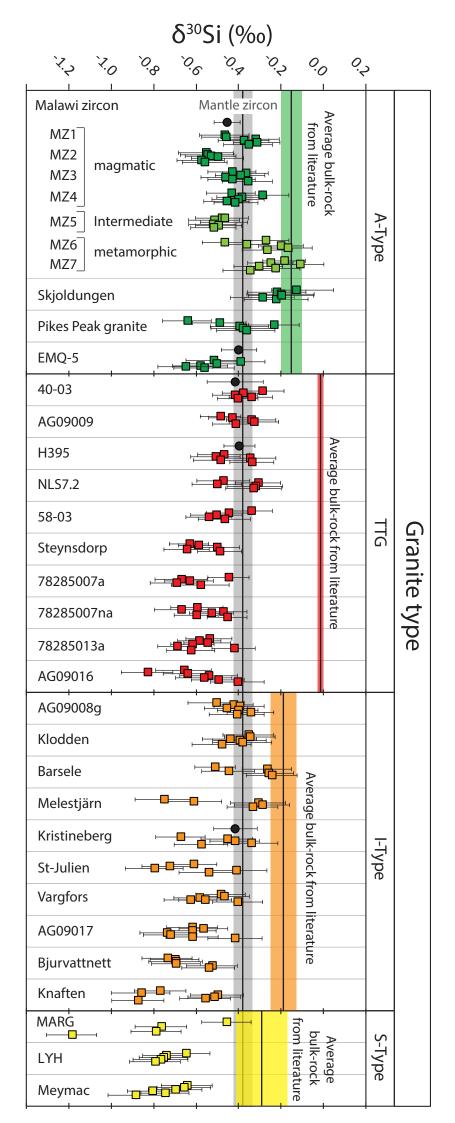
Figure 2. Chondrite-normalized (McDonough and Sun, 1995) multi-element plots for granitic zircons. Color-coding is the same as in Figure 1. Data for AS3 zircon are from this study and given for comparison.

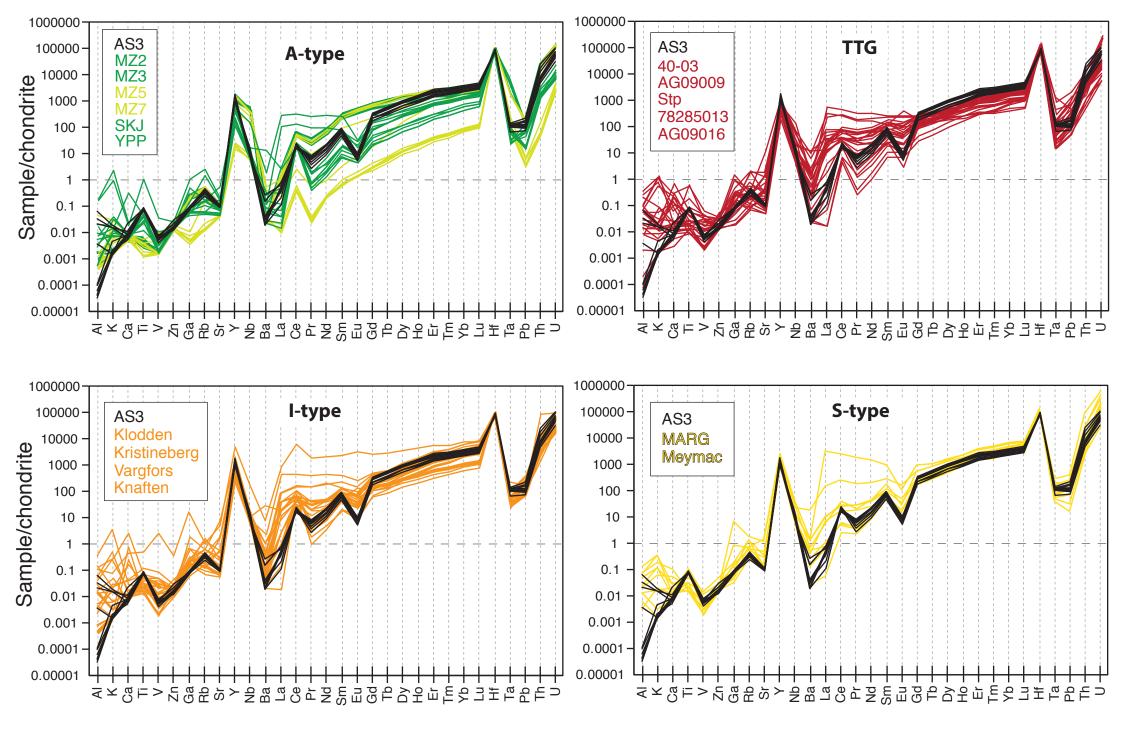
Figure 3. Zircon reference AS3-normalized multi-element plots for granitic zircons. Color-coding is the same as in Figure 1. Data for AS3 zircon are from this study (Table S4) and given for comparison. The AS3 elemental values used to normalize zircon data are averages of measurements from this study for which Al concentration was below 1 ppm in order to obtain a least altered and representative igneous composition. Figure 4. Back-scattered electron (BSE) image of the megacrystic zircon from Malawi with details about *in-situ* measurement locations (MZ1-MZ7). Also indicated are isotopic and chemical information obtained from these zones. Note the general fine oscillatory zoning from core to rim typical of magmatic zircons on which is superimposed dark and homogeneous metamorphic zones together with fractures and mid-grey patchy domains. Red symbols in Figures B-G correspond to analyses done within magmatic domains visible in Figure A, whereas grey symbols are from metamorphic domains and blue symbols from "intermediate" patchy domains. See text for details.

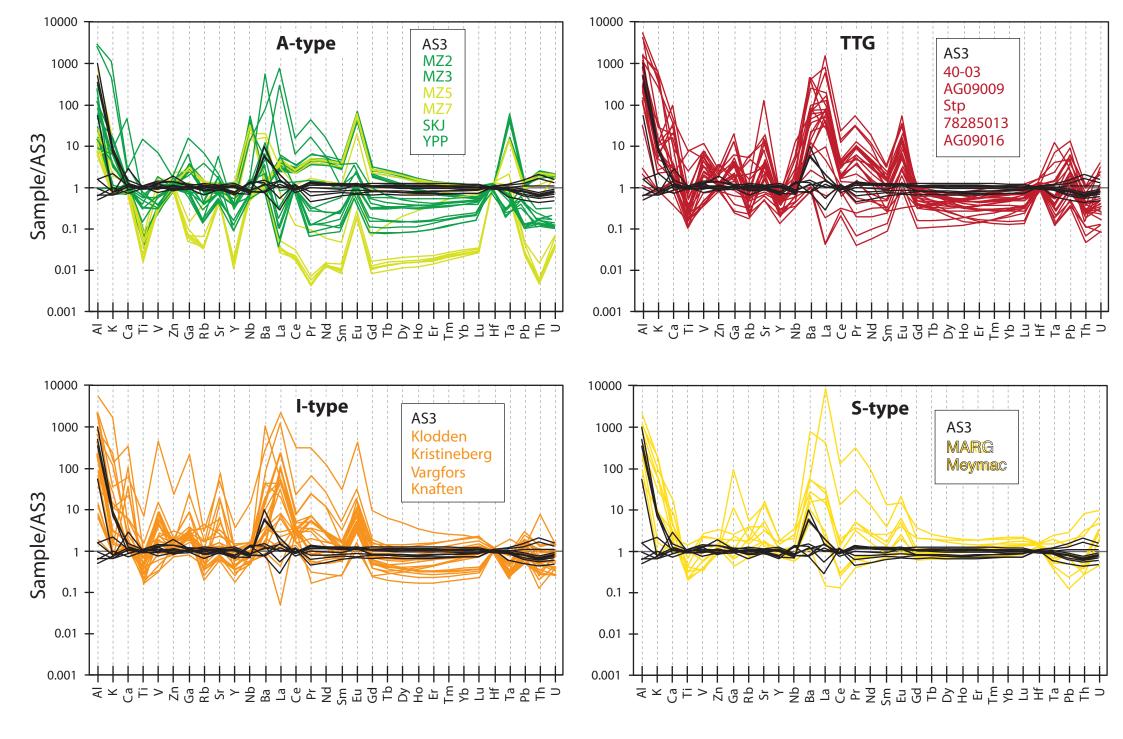
Figure 5. Composite figure that presents zircon δ^{30} Si and chemical data from TTGs (A), from I-type granite (B) and (C), and from S-type granite (D). Figure (A) shows that metamorphism, which is marked by a decrease of Th/U down to < 0.2, lowers zircon δ^{30} Si. Note that AG09-016 zircons with the highest δ^{30} Si value, which also have typical igneous Th/U, fall in the range of δ^{30} Si values exhibited by "pristine" zircons from other TTGs. Figure (B) shows that high Ca concentration in zircon, which is a sign of alteration/weathering, can possibly drag δ^{30} Si down. Ti concentration in the same altered zircon has consequently been increased (Figure C), hence, compromising the meaningfulness of determined Ti-in-zircon temperature. Finally, Figure D presents the only relationship we obtained in our dataset between δ^{30} Si and Ti-in-zircon temperature. As illustrated by the arrows, this relationship cannot solely be explained by temperature decrease but has to involve a change in SiO₂, and perhaps a source control.

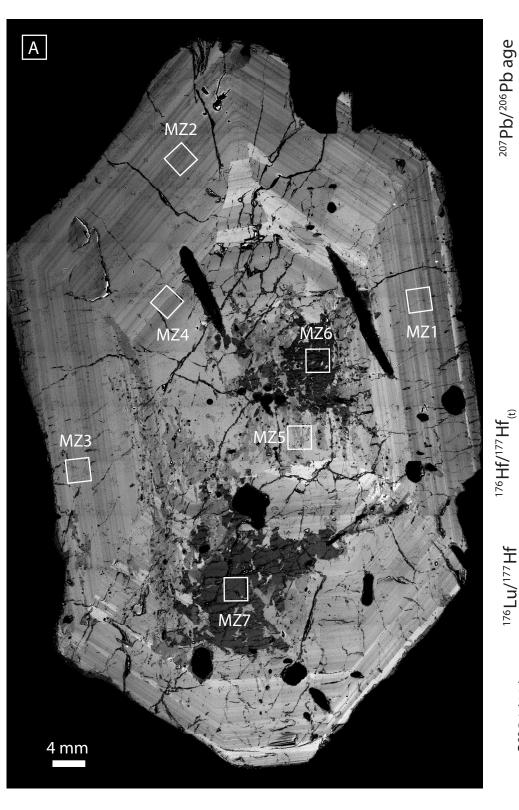
Figure 6. Silicon isotope fractionation between zircon and melt, expressed as Δ^{30} Si_{WR-} $_{Zrc}$, as a function of Ti-in-zircon temperature (T°C) and $10^6/T^2$ (in Kelvin). We only plotted samples for which we analyzed trace elements and that were not affected by secondary processes (see text fo details). Also indicated are SiO₂ contents of corresponding whole-rock (WR) samples (Table 3). The solid and dashed lines with associated SiO₂ contents are from Trail et al. (2019). Δ^{30} Si_{WR-Zrc} were calculated assuming bulk-rock A-type granite δ^{30} Si of -0.15‰, TTG δ^{30} Si of -0.0‰, I-type δ^{30} Si of -0.20‰, and S-type granite δ^{30} Si of -0.30‰ (Deng et al., 2019). Note the general good consistency between zircon group positions and WR SiO₂ contents for samples with simple distribution (e.g., $SKJ = 58 \text{ wt}\% \text{ SiO}_2$; Klodden = 58 wt% SiO_2 ; 40-03 = 71 wt% SiO₂; AG09-009 = 72 wt% SiO₂; Meymac = 70 wt% SiO₂). Also note the spread in Δ^{30} Si_{WR-Zrc} values for other samples (e.g., Pikes Peak granite, Knaftern, MARG) that indicate heterogeneous SiO₂ that most likely arise from fractional crystallization, magma mingling, and/or change in the source of parental magma, or crustal assimilation. In fact, clay-rich sediments can have δ^{30} Si down to approximately -2‰, thereby artificially increasing the Δ^{30} Si_{WR-Zrc} (Ziegler et al., 2005; Opfergelt et al., 2012; Trail et al., 2018). Color-coding is the same as in Figure 1.

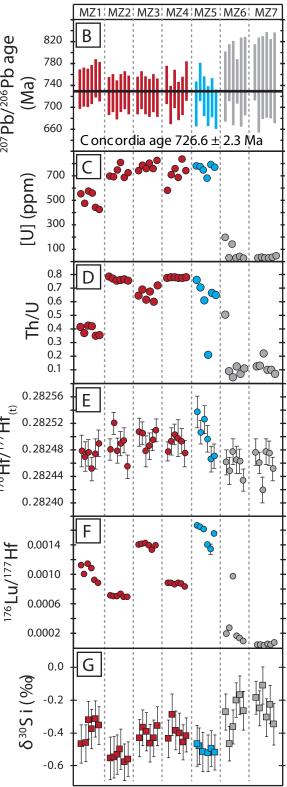
Figure 7. Silicon isotope signatures of analyzed zircons filtered for secondary processes and grouped according to major granite types. Color-coding is the same as in Figure 1. Each circle corresponds to the average zircon δ^{30} Si obtained from one sample. Also indicated are the values for the various granite types (Savage et al., 2012; Poitrasson and Zambardi, 2015; Deng et al., 2019) and the within-run precision of LA-MC-ICP-MS analyses. Note the similar pattern between zircon and whole-rock δ^{30} Si. Figure 8. Binary plot showing the relationship between A value and whole-rock SiO₂ content (A) and its consequence for the Δ^{30} Si_{WR-Zre} versus WR SiO₂ plot (B) used for interpretations of zircon δ^{30} Si and Ti-in-zircon temperature. Color-coding in (A) is the same as in Figure 1. Numbers next to lines are respective A values in (B). Also reported in (B) are the slopes and corresponding SiO₂ contents from Qin et al. (2016) and Trail et al. (2019). Compared to this study, slopes (i.e., A values) derived from theory (Qin et al., 2016) seem to under-estimate Si isotope fractionation between zircon and high-SiO₂ melts, whereas slopes from experiments tend to under-estimate Si isotope fractionation between zircon and low-SiO₂ melts (Trail et al., 2019). However, note that uncertainties on A values are similar to those on δ^{30} Si, and hence on Δ^{30} Si_{WR-Zre}, that vary between 0.1 and 0.2 (2SD absolute values) which is not shown in (B). Moreover, the fit in (A) is linear but the actual relationship between A value and SiO₂ may not be linear as noted in Trail et al. (2019) and could instead require a polynomial fit such as the modeling of fractionation behavior done in Qin et al., (2016).

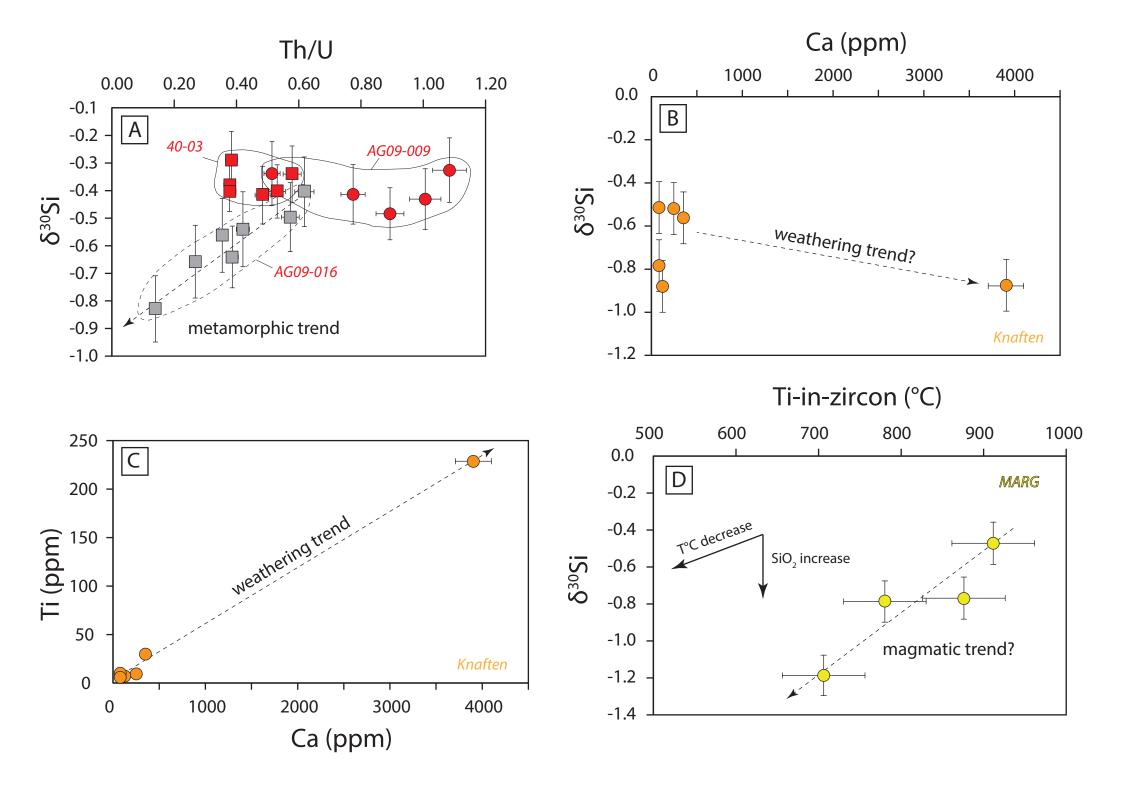


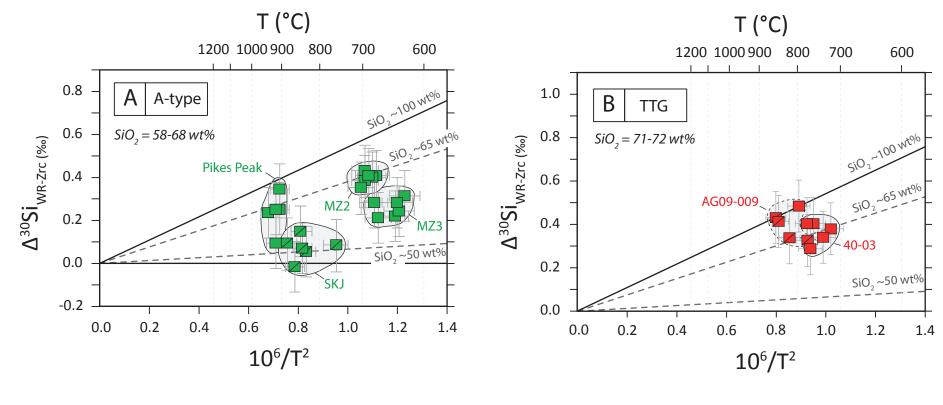












5102-100 Wt% .

510

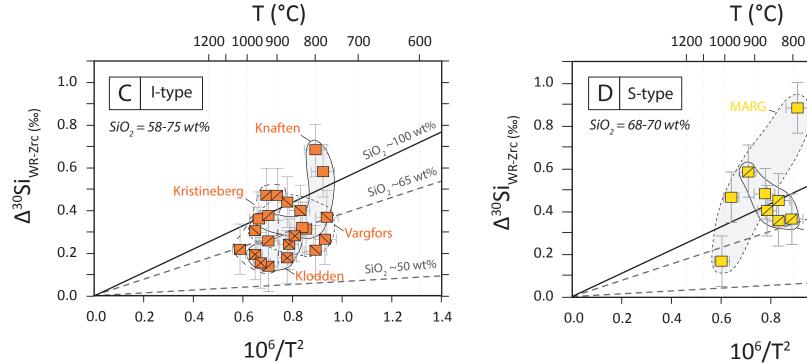
1.0

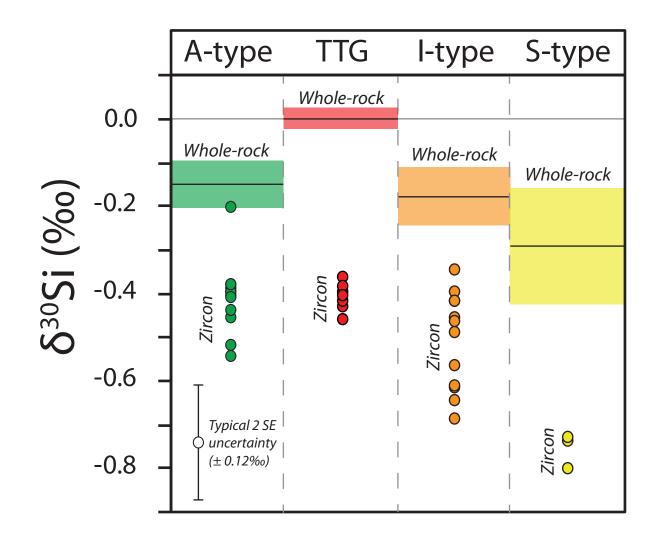
~65 W

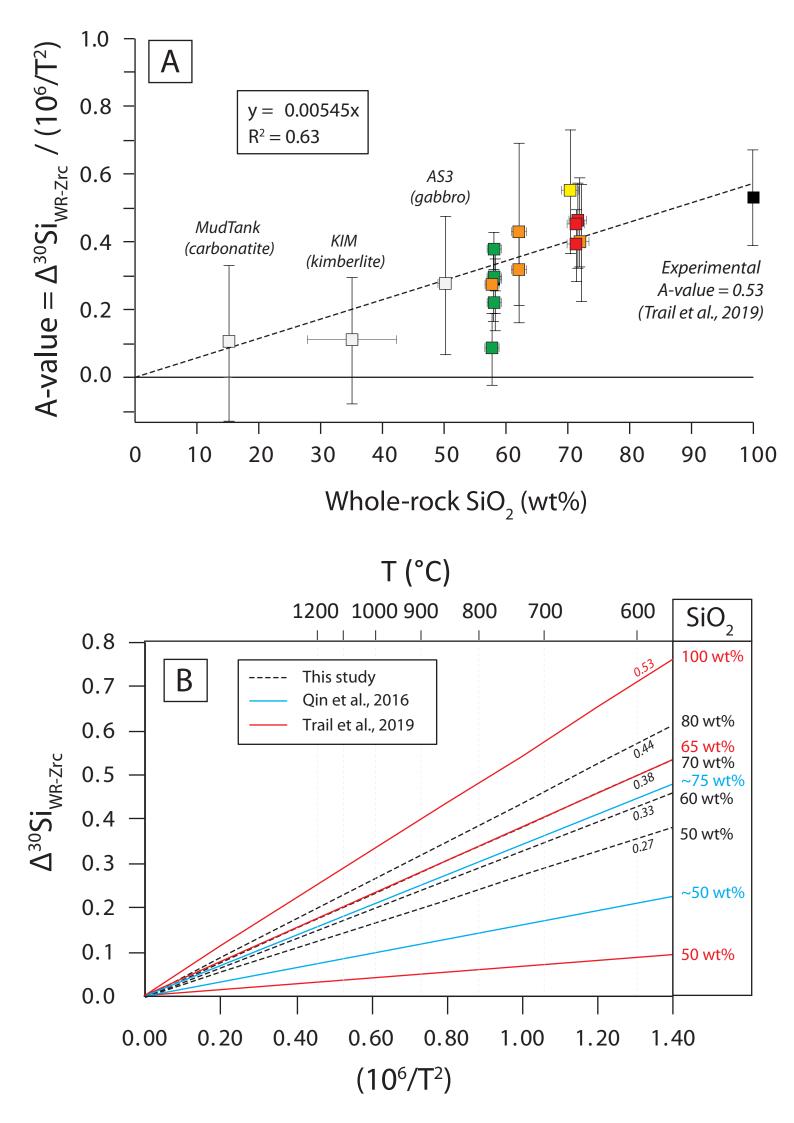
 $SiO_2 \sim 50 \text{ wt}\%$

1.2

1.4







Supplementary Material

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Stable isotope geochemistry of silicon in granitic zircon

<u>Highlights :</u>

- Zircons from various granite types were analyzed for Si isotopes
- Si isotopes were measured using laser-ablation and solution MC-ICP-MS
- Metamorphism and alteration can modify zircon Si isotope composition
- Si isotope fractionation in zircon is controlled by temperature and melt SiO₂ wt%
- Si isotopes in zircon can discriminate granite types and track magma evolution

Declaration of interests

 \boxtimes The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

□The authors declare the following financial interests/personal relationships which may be considered as potential competing interests: