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1	Simultaneous measurements of electrical conductivity and seismic velocity of partially					
2	molten geological materials: Effect of evolving melt texture					
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11	4081)					
12						
13	Key Points:					
14	1. Electrical conductivity increases with evolving melt texture; transient melt textures can					
15	underestimate the electrical conductivity.					
16	2. Acoustic velocities are not strongly affected by evolving melt texture for highly wetting					
17	melt					
18	3. Acoustic velocities measurements are more appropriate for estimating the melt fraction.					

19 Abstract

Comparison between geophysical observations and laboratory measurements yields contradicting 20 estimations of the melt fraction for the partially molten regions of the Earth, highlighting 21 22 potential disagreements between laboratory-based electrical conductivity and seismic wave velocity measurement techniques. In this study, we performed simultaneous acoustic wave 23 velocity and electrical conductivity measurements on a simplified partial melt analogue (olivine 24 + mid oceanic ridge basalt, MORB) at 2.5 GPa and up to 1650 K. We aim to investigate the 25 effect of ongoing textural modification of partially molten peridotite analog on both electrical 26 conductivity and sound wave velocity. Acoustic wave velocity (Vp and Vs) and EC are measured 27 on an identical sample presenting the same melt texture, temperature gradient, stress field and 28 chemical impurities. We observe a sharp decrease of acoustic wave velocities and increase of 29 electrical conductivity in response to melting of MORB component. At constant temperature of 30 31 1650 K, electrical conductivity gradually increases, whereas acoustic velocities remain relatively constant. While the total MORB components melt instantaneously above the melting 32 temperature, the melt interconnectivity and the melt distribution should evolve with time, 33 affecting the electrical conduction. Consequently, our experimental observations suggest that 34 acoustic velocities respond spontaneously to the melt volume fraction for melt with high wetting 35 properties, whereas electrical conduction is significantly affected by subsequent melt texture 36 modifications. We find that acoustic velocity measurements are thus better suited to the 37 determination of the melt fraction of a partially molten sample at the laboratory time scale 38 (~hours). Based on our estimations, the reduced Vs velocity in the major part of the low velocity 39 zone (LVZ) away from spreading ridges can be explained by 0.3 to 0.8 vol. % volatile-bearing 40

- 41 melt and the high Vp/Vs ratio obtained for these melt fractions (1.82-1.87) are compatible with
 42 geophysical observations.
- 43 Keywords: Electrical Conductivity, Acoustic Wave Velocity, Low Velocity Zone, Dihedral
 44 angle, Melt fraction, MORB.
- 45

46 1. Introduction

The Earth's asthenosphere is characterized by a region of high electrical conductivity (> 47 0.05 S/m) [Shankland and Waff, 1977], ~3-8 % reduction of acoustic wave velocity and high 48 seismic attenuation [Anderson and Sammis, 1970; Romanowicz, 1995]. A low degree of partial 49 melting has often been considered as a viable explanation (partial melting hypothesis), because 50 the magnitude of seismic and electrical anomalies cannot be explained by the temperature effect 51 alone [Fischer et al., 2010]. Alternative mechanisms based on solid state processes, such as 52 anelastic relaxation [Goetze, 1977; Stixrude and Lithgow-Bertelloni, 2005] and hydrogen 53 54 diffusion [Karato, 1990] in mantle minerals have also been proposed (null hypothesis). However, the recent finding of young alkali basalt (< 10 Ma) on the 135 million-year-old Pacific Plate 55 [Hirano et al., 2006] provides strong physical evidence for partial melting at the top of the 56 asthenosphere. 57

The criteria for melting in the asthenosphere have been discussed in a number of recent 58 papers [Galer and O'Nions, 1986; Plank and Langmuir, 1992; Dasgupta and Hirschmann, 59 2006]. Volatile-assisted melting in the asthenosphere is favored as the mantle temperatures at the 60 relevant depths are expected to be lower than the dry peridotite solidus [Dasgupta and 61 Hirschmann, 2006]. The volatile contents of the primitive mantle samples suggest mantle 62 abundances of ~ 150 wt. ppm of H₂O and ~ 100 wt. ppm of CO₂ [Saal et al., 2002], while CO₂ 63 contents of up to 1800 wt. ppm have been reported in undegased sources [Cartigny et al., 2008]. 64 65 The recent discovery of young alkali basalt associated with volcanism along fractures in the lithosphere indicates up to 5 wt. % CO₂ and 1.0 wt. % of H₂O volatile contents [Okumura and 66 Hirano, 2013]. However, the measurements based on melt inclusions in minerals and quenched 67 glasses indicate a global average of about 3000 wt. ppm of H₂O and 170 wt. ppm of CO₂ in 68

natural MORB [*Naumov et al.*, 2014]. A substantial contribution of volatiles to the melting can
be expected at low temperature regions in the asthenosphere [*Sifré et al.*, 2014; *Yoshino et al.*,
2010].

The reduced seismic velocity and elevated electrical conductivity have been widely used 72 as evidence for the presence of melt in the Earth's interior [Anderson and Sammis, 1970]. The 73 magnitudes of the seismic velocity and conductivity variations are directly linked to the melt 74 fraction, therefore comparison of geophysical data with laboratory models has long been 75 considered as the most plausible way to quantify the melt contents in partially molten regions of 76 77 the Earth [Anderson and Sammis, 1970; Shankland and Waff, 1977]. The accurate determination of melt volume fraction in the asthenosphere is a key constraint for the plate tectonics and mantle 78 convection models [Schmerr, 2012]. Apart from identifying the partially molten regions and 79 80 quantifying the melt fractions, the seismic and electrical methods can also be used to characterize their spatial distribution. For example, laboratory-based experiments [Caricchi et al., 2011; 81 Zhang et al., 2014; Pommier et al., 2015] have been able to attribute the seismic and electrical 82 anisotropies observed at spreading ridge environments to the shear localization of melt due to 83 plate motion. 84

The presence of a melt significantly modifies the viscoelastic properties of mineral assemblages. The critical parameters are the volume fraction and the melt microstructures [*Kohlstedt*, 1992]. Unfortunately, experimental determinations of seismic velocity on realistic melt compositions are limited to a few studies. An early measurement of Vp and Vs in a meltbearing peridotite reported no significant effect of melt fractions below 3.0 vol. % [*Sato et al.*, 1989]. The measurements based on torsional forced oscillation of melt-bearing olivine indicate reduced seismic velocities, and high attenuation can be observed for melt fractions as low as 0.01

92 vol.% [Faul et al., 2004], suggesting a possible melt fraction of 0.1 to 1 vol. %, for the average grain size variation in the upper mantle from 1 and 10 mm, respectively. The recent experimental 93 developments allow accurate determination of seismic velocity measurements of partially molten 94 rocks at the pressure and temperature conditions expected at the Earth's interior [Chantel et al., 95 2016] and predicted about 0.2 vol. % melt content in the asthenosphere. On the contrary, the melt 96 97 fraction estimations based on the acoustic velocity of analogue systems [Takei, 2000] indicate significantly higher melt fractions than those predicted using realistic upper mantle melts [Faul 98 et al., 2004; Chantel et al., 2016]. For example, the 6.6 % melt required to explain 10 % Vs 99 100 reduction in Borneol-diphenylamin analogue system [Takei, 2000] is considerably higher than 101 the about 1 % melt required by basaltic melt to explain a similar velocity reduction [Chantel et al., 2016]. 102

103 The dependence of acoustic wave velocities and attenuation upon melt fraction and grainscale melt distribution has also been discussed in several theoretical studies. These studies were 104 based on ideal melt geometries and explained using the oblate spheroid model [Schmeling, 105 1986], tube model [Mavko, 1980], the crack model [O'Connell and Budiansky, 1974] and models 106 based on grain boundary wetness or "contiguity" [Takei, 1998, 2002; Yoshino et al., 2005; Hier-107 108 Majumder, 2008]. The calculations based on the finite element method on melt geometries led [Hammond and Humphreys, 2000] to conclude more than 1 % melt is required to explain 3.6 % 109 and 7.9 % velocity reduction for Vp and Vs respectively, which is significantly lower than the 110 111 6.2 % and 11 % reductions observed in laboratory measurements [Chantel et al., 2016]. The naturally occurring, randomly distributed melt [Faul et al., 2004; Chantel et al., 2016] is shown 112 to have a significant effect on seismic velocity compared to the simplified melt textures assumed 113 114 in theoretical models [Hammond and Humphreys, 2000; Takei, 2002; Yoshino et al., 2005] or analogue systems [*Takei*, 2000]. The considerably higher melt volume fractions required in
theoretical models can be attributed to the idealized geometries, such as planar cracks, spheres,
ellipsoids, or simplified cuspate forms, which may not represent the true melt geometries in
naturally occurring melt [*Kohlstedt*, 1992; *Faul et al.*, 1994; *Hammond and Humphreys*, 2000],
limiting their applications to partially molten regions in the asthenosphere.

120 An early electrical conductivity measurement on basaltic melt suggested 5-10 vol. % melt needed to account for the observed electrical anomalies in the asthenosphere [Tyburczy and Waff, 121 1983]. Similar amount of basaltic melt (5%) also provided compatible values to geophysical 122 123 observables with values up to 0.1 S/m [Maumus et al., 2005]. However, recent studies suggest the presence of much lower volume fractions; 0.3-3.0 vol. % for hydrous basaltic melt [Yoshino 124 et al., 2010; Ni et al., 2011], or less than 0.3 vol. % for carbonatitic melt [Gaillard et al., 2008; 125 *Yoshino et al.*, 2010]. The electrical conductivity of volatile enriched basalt (15–35 wt.% CO_2) 126 and about 2-3 wt.% H₂O) indicates about 0.1-0.15 vol.% melt could explain the observed 127 conductivity anomalies [Sifré et al., 2014]. The development of melt interconnectivity in 128 partially molten rocks is known to have a profound effect on electrical conductivity [*Waff*, 1974, 129 Maumus et al., 2005]. However, the number of studies investigating the influence of melt 130 131 microstructures on EC is extremely limited. The study by [ten Grotenhuis et al., 2005] showed that a melt geometry evolving from isolated triple junction tubes at 0.01 % of melt to a network 132 of interconnected grain boundary melt layers at 0.1% of melt has a greater effect on electrical 133 134 conductivity.

Various other experimental techniques have been used to constrain the melt fraction associated with the LVZ in the Earth's asthenosphere. Geochemical constraints from trace elements partitioning suggest that low degree of melting (less than 1 %) can be generated at

greater depth (below than 100 km) [*Salters and Hart, 1989*]. Similarly, studies based on
experimental petrology, such as volatile (H₂O and CO₂) effect on peridotite solidus, indicate the
melt fraction in the asthenosphere LVZ has to be 0.1 vol. % or less [*Plank and Langmuir,* 1992, *Dasgupta and Hirschmann,* 2007].

Both experimental and theoretical models acknowledge that the volume fraction and 142 143 spatial distribution of melt play an integral part in modifying the seismic and electrical properties of partially molten rocks. However, large discrepancies still remain in the laboratory estimations 144 (regardless of the technique) of the amount of melt volume fraction present in the asthenosphere 145 [Pommier and Garnero, 2014; Karato, 2014]. Due to the large number of studies addressing the 146 electrical properties of melt, the disagreement between laboratory-based electrical conductivity 147 measurements is highly visible [Karato, 2014]. The influence of volatile contents could be one of 148 the key parameters controlling the conductivity of the resulting melt [Yoshino et al., 2010; Ni et 149 al., 2011; Sifré et al., 2014]. A model based on chemical variation in the melt has been proposed 150 to explain the apparent disagreement on melt fraction estimations between electrical conductivity 151 measurements and seismic models [Pommier and Garnero, 2014]. The discrepancy may 152 primarily stem from the absence of systematic experimental investigation into the structural 153 154 factors influencing the EC in partially molten systems. The effect of melt fraction on seismic velocity has been mostly limited to numerical models. Significant disagreement between these 155 theoretical models is still present due to the choice of melt geometries [Yoshino et al., 2005]. A 156 157 comparison with recent laboratory-based seismic velocity measurements on realistic partially molten materials [Chantel et al., 2016] indicates a significant underestimation of seismic 158 response by theoretical models [Takei, 2000; Yoshino et al., 2005]. The cross-correlation of melt 159

160 fraction estimations based on theoretical seismic models and laboratory electrical conductivity is161 at present a highly uncertain exercise.

The effect of melt texture and chemical compositions (volatiles) have long been assumed for the observed inconsistency, but there has never been a systematic study on how the evolving melt textures influence the electrical conductivity and seismic velocity (SV). Similar, melt textures and melt contents during high pressure, high temperature experiments are strongly affected by the stress field and the temperature distribution within the sample and it is highly unlikely that two experiments would yield identical melt distributions.

168 In this study, we aim to investigate the effect of ongoing textural modification of partially molten peridotite analog on both electrical conductivity and sound wave velocity. Here we have 169 170 developed a novel high-pressure multi-anvil cell design to investigate simultaneously the seismic 171 and electrical properties of partially molten samples. Acoustic wave velocity (Vp and Vs) and EC are measured on an identical sample presenting the same temperature gradient, stress field and 172 the chemical impurities, which all influence the melt content and the melt texture in partially 173 molten high pressure samples. This critical improvement enables us to compare the seismic and 174 electrical responses to the onset of melting, to different melt volume fractions and to the 175 evolution of melt interconnectivity of a partially molten sample. Based on our observations, we 176 suggest possible scenario which may resolve the observed discrepancy of melt fraction 177 estimations based on EC and SV measurements. 178

179

180 **2. Methods**

181 **2.1 Sample preparation**

Samples used in this study were a powder mixture of natural San Carlos (SC) olivine and 182 volatile-rich natural MORB glass (location 6°44'N, 102°36'W, collected during the Searise-1 183 research cruise). The volatile content is estimated to be 2730 (± 140) ppm wt. H₂O and 165 (± 40) 184 185 ppm wt. CO₂ [Andrault et al., 2014], comparable with the average H₂O and CO₂ levels observed in MORB from diverse geological settings [Naumov et al., 2014]. The MORB glass and 186 inclusion free, hand-picked, SC olivine crystals were crushed separately and reduced to fine 187 grain powders (see grain size distribution in figure S1). The water content analysis of the San 188 Carlos olivine indicates less than 1 wt. ppm of water [Soustelle and Manthilake, 2017]. These 189 powders were then mixed in predetermined weight proportions to obtain the desired melt 190 fractions at high temperature. The accurate determination of melt fraction using the image 191 analysis is an uncertain exercise. The mixing of MORB with olivine results in an accurate 192 control of the melt fraction in the sample as the MORB component melts instantaneously above 193 its melting temperature, which is significantly lower than the olivine solidus. This procedure has 194 been extensively used to obtain a controlled melt fraction in high-pressure experiments [Faul et 195 196 al., 1994; Cmíral et al., 1998; Maumus et al. 2005; Yoshino et al., 2010; Caricchi et al., 2011; Zhang et al., 2014; Chantel et al., 2016]. While this technique is suitable for obtaining controlled 197 melt fractions (nominal melt fractions) in laboratory experiments, it cannot be used as a 198 199 substitute for the physical property measurements of incipient melting scenarios [Sifré et al., 2014]. We prepared different starting materials with MORB volume fractions of 0.1, 0.5, 1 and 2 200 vol. % mixed with San Carlos olivine. The powder mixtures were ground with an automatic 201 202 agate mortar for more than 2 hours to obtain a homogeneous distribution of the MORB

203 component. The starting powder average grain size was estimated to be $3.74 \pm 3.32 \ \mu m$ (Fig. 204 S1). In order to achieve high accuracy during weighting the powder, we prepare more than 5 g 205 for each composition. The resulting powder mixtures were then hot pressed at 2.5 GPa and 1100 206 K for 2 hours using a 1500 ton multi-anvil apparatus. The low temperature for hot pressing 207 experiments (below the melting temperature of MORB) ensures that the starting materials are 208 melt free and thus that the evolution of melt texture occurs during the conductivity and velocity 209 measurements.

210

211 2.2 High-pressure high-temperature experiments

High-pressure, high-temperature experiments were performed using a 1500 ton Kawai-212 type multi-anvil apparatus at Laboratoire Magmas et Volcans, Clermont-Ferrand, France. For 213 experiments conducted at 2.5 GPa, we used octahedral pressure media composed of MgO and 214 Cr₂O₃ (5 wt. %) in an 18/11 multi-anvil configuration (octahedron edge length / anvil truncation 215 edge length) (Fig. 1). The assembly was designed to accommodate the geometrical requirements 216 for measurements of Vp, Vs and EC in a single high pressure cell. The pre-synthesized 217 cylindrical sample was inserted into a hexagonal boron nitride (hBN) capsule. The use of high-218 219 purity hBN sintered at high temperature and pressure without binder (Type - BN HP, FINAL Advanced Materials) prevents the B_2O_3 reacting with the silicate melt. The hBN capsule also 220 helps to electrically insulate the sample with respect to the furnace. The furnace is composed of a 221 222 50 µm thick cylindrical Re foil, with apertures for the electrode and the thermocouples wires. A zirconia sleeve was placed around the furnace to act as a thermal insulator. Oxygen fugacity of 223 the sample was not controlled during the experiments, but should be below Re-ReO₂ buffer. 224

We placed two electrodes made of Re discs (25 μ m thick) at the top and bottom of the 225 cylindrical sample. A tungsten-rhenium (W₉₅Re₅-W₇₄Re₂₆) thermocouple junction was placed at 226 one end of the sample to monitor the temperature. On the opposite side it was connected to a 227 single $W_{95}Re_5$ wire (See Fig. S2 for details on electrode connection). We collected impedance 228 spectra between the two W₉₅Re₅ wires. Cylindrical MgO ceramic sleeves were used to insulate 229 230 the electrode wires from the furnace. A dense Al_2O_3 buffer rod was placed between one of the tungsten carbide (WC) anvil truncations and the sample to enhance the propagation of elastic 231 waves and to provide sufficient impedance contrast to reflect ultrasonic waves at the buffer rod-232 233 sample interface. Both ends of the anvil, the alumina buffer rod and the samples were mirror polished using 0.25 µm diamond pastes in order to enhance mechanical contacts. All ceramic 234 parts of the cell assembly, including the pressure media, were fired at 1373 K prior to the 235 assembling in order to remove any absorbed moisture. 236

237

238 2.3 Acoustic wave velocity measurements

Acoustic wave velocities of samples were measured using the ultrasonic interferometry 239 technique [Chantel et al., 2016]. In this method, electrical sine wave signals of 20–50 MHz (3–5 240 cycles) with $V_{peak-to-peak}$ of 1–5 V were generated by an arbitrary waveform generator (Tektronix 241 AFG3101C) and were converted to primary (V_P) and secondary (V_S) waves by a 10° Y-cut 242 LiNbO₃ piezoelectric transducer attached to the mirror polished truncated corner of a WC anvil. 243 244 The resonant frequency of the transducer is 50 MHz for compressional waves (*P*-waves) and 30 MHz for shear waves (S-waves). Elastic waves propagated through the anvil, the alumina buffer 245 rod (BR) and the samples, and were reflected back at the anvil-BR, BR-sample, and sample-246 247 electrode interfaces. We also consider possible reflections from the Re electrodes [Davies and

O'Connell, 1977; Jackson et al., 1981; Niesler and Jackson, 1989] (Text S1 c). The reflected elastic waves were converted back to electrical signals by the transducer and captured by a Tektronix DPO 5140 Digital Phosphor Oscilloscope at a rate of 5×10^9 sample/s. Signals at 20, 30, 40 and 50 MHz were recorded at each temperature step. The two-way travel time for the acoustic waves propagating through the sample can be determined by the time difference between the arrivals of the echoes from the BR-sample interface and the sample-electrode interface by the pulse-echo overlap method [*Kono et al.*, 2012].

255

256 2.4 Electrical conductivity measurements

EC measurements were performed using the ModuLab MTS Impedance/Gain-phase 257 analyzer in the frequency range of 10^6 - 10^1 Hz. Polyphasic samples are characterized by a 258 combination of resister-capacitor/constant phase element (R-C/CPE) circuits and the resistance 259 can be obtained by fitting the impedance spectra to appropriate equivalent circuits (Fig. S3). 260 Once the sample resistance has been determined, conductivity can be calculated using the sample 261 dimensions determined at each temperature using the thermal expansion of the constituent 262 phases. The insulation resistance of the assembly was determined in a preliminary experiment 263 using an hBN rod at similar pressure-temperature conditions and was observed to be lower than 264 the sample resistance. 265

At the target pressure of 2.5 GPa, the sample was kept at 500 K for more than 12 hours. While maintaining 500 K, the electrical resistance of the samples, measured at regular intervals, usually increases due to the removal of the moisture absorbed by the sample and surrounding materials. This step is crucial to prevent the moisture (H₂O) being incorporated into the sample at higher temperatures [*Manthilake et al.*, 2009]. The next heating cycle started once the resistance

271 reached a steady value, which is often 1-2 orders of magnitude higher than the resistance measured at the beginning of the heating cycle. We generally performed several heating-cooling 272 cycles at temperature steps of 50-100 K, until sample resistance was reproducible between the 273 heating and cooling paths. This procedure minimizes the uncertainty of EC measurements due to 274 impurities (H₂O and CO₂). Once the solid sample EC was reproducible (without moisture), the 275 276 temperature was gradually increased in smaller temperature steps (25 K) to initiate melting. Sample melting is characterized by a drastic decrease in the sample resistance (increase in 277 conductivity). Finally, the temperature was kept constant at 1650 K and impedance spectra were 278 279 collected at regular intervals for more than 1 hour.

280

281 **2.5 Melt textures and dihedral angle measurements**

Micro-textures of the recovered samples were investigated with a Field Emission Gun 282 Scanning Electron Microscope (FEG-SEM) with an accelerating voltage of 15 kV and working 283 distance of 9 - 11.6 mm. High magnification back scattered electron (BSE) images were obtained 284 in order to identify the degree of interconnectivity and the structure of the melt at the grain 285 boundaries in partially molten samples after SV and EC measurements. The presence of the hard 286 287 alumina piston may introduce differential stresses to the sample, resulting in a shape-preferred orientation (SPO) in partially molten samples [Bussod and Christie, 1991]. To characterize the 288 possible melt alignment in an olivine matrix, we performed image analyses on BSE images along 289 290 a section parallel to the axis of the cylindrical sample. The orientation of the long axis of melt pockets and area of the melt pockets were obtained by image processing techniques using Matlab 291 software (Fig. S4). 292

294 **2.6 Grain size and grain orientation distribution**

The grain size of our samples was estimated using two different techniques: the intercept 295 method, and using FOAMS software [Shea et al., 2010]. The intercept method estimates the 296 number of intersections of grain boundaries with a random line drawn across the sample. The 297 length of the line is important in order to statistically cross enough grains. The FOAMS software 298 measures every isolated particle from skeletonized images and estimates its morphological 299 parameters: area, perimeter, shape from 2D ellipse with a long and short axis, etc. The program 300 also calculates 2D parameters such as aspect ratio and elongations. From the binary images, the 301 302 code can convert 2D morphological information into 3D information using the equivalent diameter for spherical geometry by means of stereological conversion equations from [Sahagian 303 and Proussevitch, 1998]. This program works properly for all type of samples for 2D 304 information. Volumetric estimations (2D to 3D) can be performed when the grains are mostly 305 rounded and do not show strongly elongated shapes. Results are given in table S1. 306

307

308 2.7 Experimental uncertainties

Experimental measurements of Vp, Vs and EC are subjected to uncertainties originating 309 310 from the estimation of temperature pressure, sample dimensions and fitting errors. Errors have been estimated to be 2.5 % for seismic velocities (2σ) , 5 % for velocities drops (2σ) and 5 % for 311 EC values (2σ). Errors on the melt fractions are less than 1% relative (ex: 1±0.01 % of MORB). 312 313 Detailed sources of uncertainties for each technique and error propagation calculations are given in supporting information (Text S1) [Bouhifd et al., 1996; Gillet et al., 1991; Li et al., 2007]. 314 Error bars are reported in each figure when larger than the symbol size except figure 4 and 8a) 315 316 for visibility.

318 **3. Results**

319 **3.1 Acoustic velocity**

The acoustic wave velocities obtained for samples containing SC olivine and 0.1, 0.5, 1 320 and 2 % nominal volume fractions of melt are shown in figure 2. Below the melting temperature, 321 Vp and Vs decrease with increasing temperature, emphasizing the characteristic decrease of bulk 322 323 and shear modules with temperature. Upon melting of the MORB component, which is at about 1590 K, both Vp and Vs decrease significantly for the samples with 0.5 to 2 % MORB. The 324 magnitude of the velocity drop is positively correlated to the MORB fraction in the sample, but 325 326 no significant change is observed at the melting temperature for pure olivine and the sample with 0.1 vol. % melt. After the initial decrease in response to the MORB melting, the acoustic 327 velocities Vp and Vs remain unchanged while maintaining the sample at a constant temperature 328 of 1650 K. 329

330

331 **3.2 Electrical conductivity**

The electrical conductivity of samples containing 0.1, 0.5 and 2 vol. % of nominal melt 332 fractions are shown in figure 3. At the melting temperature of MORB (~1590 K), the samples 333 334 with 0.5 to 2 vol. % melt indicate sudden increases in conductivity (up to a factor of 5), compared to their solid counterparts. However, no immediate change in conductivity is observed 335 for 0.1 vol. % melt upon crossing the temperature threshold. EC of all melt-bearing samples 336 337 continues to increase after the melting event, while being kept at a constant temperature of 1650 K. The rate of increase of conductivity gradually decreases with time (Fig. 3), probably 338 approaching a steady-state with time, however these 1h duration experiments didn't reached a 339 340 stable EC value over time. The conductivity values after being kept at 1650 K for more than 30

minutes indicate an increase in conductivity of 0.6 log units, a factor of 3.98, for the 0.1 % melt
sample compare to the value obtained before melting. The conductivity variations after being
kept for more than 50 minutes at 1650 K, are about 0.6 log units, a factor of 3.98, for the 0.5 %,
and a 0.4 log units, a factor of 2.51, for the 2 % melt-bearing samples.

345

346 **3.3 Textural analyses of samples and melt**

The presence of melt is clearly visible for all samples, with melt distributing along the grains boundaries as well as triple junction tubes. Interconnected melt networks are visible over a large part of each sample including the samples with 0.1 vol. % of melt (Fig. 4). Using the high resolution SEM images, we determined the wetting angles of the melt-solid interfaces, which indicate a median angle of $27 \pm 4^{\circ}$.

Table S1 presents the grain size and grain orientation parameters derived from both 352 intercept and FOAMS software, samples average grain size are similar and between 7 to 15 353 micrometres. The eccentricity is calculated from the best fitting ellipse foci and circle centre. 354 This parameter indicates how far the best fitting ellipse deviates from perfect circularity, the 355 values from 0.76 to 0.84 in our samples indicate that grains are mainly rounded but not perfect 356 spheres. Elongation parameter is expressed by $\varepsilon = (a-b)/(a+b)$, which characterizes the difference 357 between the long (a) and short (b) axes of the fitting ellipse; large values (close to 1) indicate 358 elongated particles. Our average values trends from 0.25 to 0.36, meaning grains have an elliptic 359 cross section which slightly deviates from circularity. Aspect ratio is expressed as A = b/a, and 360 characterizes the shape of the particle; large aspect ratios (close to 1) indicate particles are 361 rounded and not elongated; our high-intermediate values are good agreement with this 362 363 observation.

The analyses based on the orientation of the long axis of melt pockets indicate random distribution of melt within the olivine matrix (Fig. S4). The associated histogram indicates no significant preferential orientation.

367

368 **4. Discussion**.

369 **4.1 Effect of evolving melt texture on acoustic wave velocity and electrical conductivity**

Upon melting, partially molten samples evolve toward textural equilibrium with time, 370 thus improving the melt interconnectivity and melt redistribution within the olivine matrix (Fig. 371 372 4). The comparison of images of samples before melting and after keeping prolonged time above the melting temperature of MORB clearly demonstrate the evolution of Ol+MORB powder 373 mixture from initial non-equilibrium state (MORB is randomly distributed) to the extensive 374 wetting of crystal faces and the smoothly curved solid-melt interfaces (Fig. S5). The textural 375 equilibrium depends on several factors such as melt fraction, melt chemistry and grain size 376 distribution [Laporte and Provost, 2000]. The melt geometries in olivine-basalt systems consist 377 of grain boundary melt layers, triple junction networks [Yoshino et al., 2005, 2009] and 378 ellipsoïdal discs [Faul et al., 1994]. The continuous increase of EC observed in our experiments 379 380 can be attributed to the gradual development of an interconnected network of melt channels, which facilitate the movement of charge carried through the melt. In contrast, acoustic wave 381 propagation in a partially molten media should be more affected by the presence of melt in its 382 383 path (volume fraction), than its fine geometrical evolution subsequent to melt interconnection. We note that the EC increased quickly in the first tens of minutes and showed a flat evolution 384 with almost flat slope after about 1 hour, indicating that the textural modifications that influence 385 386 the interconnectivity of the melt can be mostly achieved within few hours. The melt takes its

final like shape very quickly (in the first hour), however complete equilibration between melt
and host olivine matrix in both chemical and textural aspects require several weeks of annealing
[*Waff and Blau, 1982; Laporte and Provost, 2000*].

390

a) Interpretation of acoustic wave velocity results

The magnitude of the drop in seismic wave velocity in response to melting is proportional 392 to the melt volume fraction in the sample (Fig. 2). Compared to the higher melt fractions, the 393 sample containing 0.1 % melt does not show abrupt variations of acoustic wave velocity in 394 395 response to the onset of melting of MORB components. This observation suggests that the volume fraction of melt has to be sufficiently large (higher than 0.1 vol. %) in order to alter the 396 seismic wave propagation through partially molten rock. Also, associated errors to seismic wave 397 velocity measurements and fitting does not allow distinguishing significant drop for low melt 398 fractions (~0.1 %). Further, the relatively constant seismic velocity at a constant temperature of 399 1650 K, after the melting of MORB, suggests that the seismic velocity is less sensitive to the 400 ongoing textural equilibration of the sample. The melt fraction in a partially molten rock with 401 complete wetting properties is observed to be the key parameter controlling the magnitude of 402 seismic velocity in geological systems. The secondary waves (Vs) are more sensitive to the 403 presence of melt due to their near zero shear modulus, which further enhances their ability to 404 detect and quantify melting in laboratory samples. 405

The comparison of present data with previous experimental and theoretical estimations of seismic velocity is shown in figure 5. While our results are consistent with that of [*Chantel et al.*, 2016], there are considerable deviations in our experimental values from those estimated based on theoretical approximations [*Takei*, 2000]. As explained previously, the disagreement may arise

410 due to the simplified melt geometries assumed in theoretical models. This observation can be further corroborated by comparing two theoretical models, one based on natural melt geometries 411 [Yoshino et al., 2005], and the other on ideal melt geometries [Takei, 2000]. The model with melt 412 arrangements similar to naturally occurring melt record a significant velocity drop for a given 413 melt fraction compared to the one assuming ideal melt distribution. However, the model based 414 on grain boundary wetness [Yoshino et al., 2005] also predicts the seismic velocities are 415 significantly affected by modifications on the pore geometry. It has been shown that the melt 416 wetting properties vary significantly with increasing pressure and volatile content [Yoshino et al., 417 418 2007]. The slight discrepancy between the present study and that of [Yoshino et al., 2005] can be explained by the change in wetting properties, due to improved melt wetting properties at high 419 420 pressure and the presence of both H₂O and CO₂ in our samples.

421

422 b) Interpretation of electrical conductivity results

The electrical conductivity variation while kept at constant temperature (at 1650 K) 423 provides valuable insights into the development of interconnected melt channels in partially 424 molten samples. For larger melt fractions (above 0.1 %) the melt network forms efficiently as 425 shown by an order of magnitude conductivity increase observed at the onset of melting. 426 However, after the onset of melting and the associated EC jump, while kept at constant 427 temperature of 1650 K, the increase in electrical conductivity for larger melt fractions (2 % with 428 429 0.4 log unit increase of EC) is smaller than the sample with a low melt fraction (0.1 % with 0.6 % melt)log unit increase of EC). This potentially indicates that when the melt fraction is sufficiently 430 large, the major portion of melt is already arranged into a well distributed network of melt 431 432 channels. On the other hand, the subsequent modifications improving the melt interconnectivity

have a significant effect on low melt fractions. It has been shown that the melt geometry in a mineral-melt aggregate is determined by the solid-solid and solid-liquid interfacial energies [*Laporte and Provost*, 2000]. The solid-liquid interfacial energies may control the interconnectivity of a partially molten medium at low melt fraction; the network of melt could be limited in its 3D extension between the olivine grains, with some surfaces remaining initially unwetted due to surface tension.

As for acoustic velocity, a sharp variation in electrical conductivity was not immediately 439 apparent for the sample with 0.1 % melt fraction. However, when maintained at 1650 K, EC 440 441 continued to increase for the 0.1 % melt sample, after 1h to 0.6 log unit higher than the conductivity of the sample before melting. This observation suggests that the electrical 442 conductivity method can be used to detect melt fractions lower than 0.1 % as long as the 443 measurements are performed on texturally equilibrated samples. As well, EC is very sensitive to 444 the onset of melting with few orders of magnitude of increase after only few minutes. Reading 445 value of sample resistance (direct measurement to infer EC) is instantaneous and can be a 446 powerful tool to detect the onset of melting during an experiment. 447

The electrical conductivity of similar olivine-basalt systems has been investigated in 448 449 previous experiments [Maumus et al., 2005, Yoshino et al., 2010; Caricchi et al., 2011; Zhang et al., 2014, Laumonier et al., 2017] (Fig. 5). While measured conductivities are located within the 450 individual EC measurements of olivine [Constable, 2006, Laumonier et al., 2017] and basaltic 451 452 melt [Presnall et al., 1972; Tyburczy and Waff, 1983; Ni et al., 2011, Laumonier et al., 2017], the partially molten systems do not display good agreement between different studies. The slightly 453 higher EC observed for partially molten samples in [Yoshino et al., 2010], compared to our study 454 455 may have been due to their use of texturally equilibrated melt-bearing samples (pre-synthesized

samples in a piston cylinder apparatus) in electrical conductivity measurements, which compare
favourably with our observations. Values comparable to "equilibrated samples" conductivities
can be retrieved, by using extrapolation of our EC versus time trend, for timescales of days or
weeks, indicating full equilibration might require might require a significant time (Fig. 5 and 6).
This again, underlines the crucial importance of the use of equilibrated EC values for safe
comparisons.

462 **4.3 The source of discrepancy**

The interpretation of seismic and electrical anomalies in terms of melt fraction often 463 results in conflicting estimations as to the extent of melting in the asthenosphere [Pommier and 464 Garnero, 2014; Karato, 2014]. A conductivity model based on major element chemistry of melt 465 attributed the apparent inconsistencies in conductivity measurements to possible chemical 466 variations in the melt [Pommier and Garnero 2014]. Their model predicts that low degree 467 melting of peridotite produces melt that is more conductive than basaltic compositions. We find 468 their approach is an important step towards unifying the seismic and electrical observations. 469 However, the melt fractions estimations used in their study were based on theoretical models, 470 which appear to underestimate the effect of melt fraction on seismic velocity. 471

Monitoring the behaviour of melt-bearing samples for an extended period of time at high pressure and high temperature remains a challenging exercise. Escape of melt during prolonged heating is one of the major sources of failure, and experimental studies often overcome this issue by shortening the duration of the *in situ* measurements at high temperature. However, our results demonstrate that the EC values can vary significantly with time within the first hour of measurements, and relatively stable EC values can be obtained once the 3D interconnected network has been established. Texturally non-equilibrium melt can lead to an underestimation of

479 the total effect on EC of a given melt fraction (Fig. 6). Comparison of such measurements with geophysical profiles, therefore, results in an overestimation of the melt fraction in the 480 corresponding region in the Earth's mantle. Values here provided after 1 h at 1650 K are not fully 481 stabilized as highlighted by the subtle slope of the fit. We also note that once the sample 482 conductivity stabilized, as a result of improved melt interconnectivity, electrical conductivity 483 values of samples containing 0.1, 0.5 and 2 % are not considerably different (less than one order 484 of magnitude). This difference becomes subtle, close to the uncertainty of measurements, for 485 higher melt fractions according to the trend shown in figure 6. This implies that uncertainties on 486 487 inferred melt fractions from EC can be very important if implied melt fractions are higher than few percents. This observation is particularly crucial for magnetotelluric (MT) profiles with low 488 spatial resolution. For these reasons, EC values here provided will not be further used for 489 geophysical implications. However, we note that the electrical conductivity measurements are 490 superior over acoustic wave velocity for detecting low melt volume fractions for samples with 491 evolved melt textures. If the wetting properties of the melt are modified by the presence of 492 significant amounts of volatiles in the melt (H₂O and CO₂) the electrical response for low melt 493 fractions is instantaneous [Sifré et al., 2014]. 494

In this study, we observe real-time Vp, Vs and EC responses during melting and consecutive textural evolution of melt. The variation of electrical conductivity subsequent to the melting of MORB can also be caused by the chemical changes occurring at high temperatures for a prolonged period of time. The effect of change in chemical composition on electrical conductivity in melt has been investigated in previous studies [*Roberts and Tyburczy*, 1999], with a general trend showing an increase in conductivity with increasing alkali and Fe+Mg contents and a decrease with increasing silica content. However, we observe that the melt composition

stays similar to the starting MORB composition during the experiments, except for a minor 502 decrease in Fe content (Table 1). The Na is an important charge carrier in silicate melt [Pfeiffer, 503 1998; Gaillard and Iacono-Marziano, 2005; Ni et al., 2011] and Na contents in our melt remains 504 similar to the starting composition. Based on the totals of chemical analyses of melt, we confirm 505 that significant volatile enrichments may not occur in our melt. Therefore, the observed 506 conductivity increase with time is not expected to be caused by any chemical modification to the 507 melt. This observation also confirms that the final melt fraction in the sample stays similar to the 508 starting material. Similarly, due to the low partition coefficient between olivine and melt 509 510 (~ 0.004) [Novella et al., 2014], the water is mostly retained by the melt phase, so proton (H⁺) diffusion in olivine affecting the electrical conductivity at high temperature can also be ruled out. 511 The constant velocity after the melting of MORB components also rules out the possible increase 512 in melt fraction in the sample at constant temperature, which is also supported by image analysis 513 and chemical mapping of the sample. Further, analyses on the orientation of the matrix and melt 514 pockets in our samples indicate random shape preferred orientation (SPO) ruling out melt 515 channelling due to possible anhydrostaticity in the high-pressure cell assembly (Fig. S4). 516

517

518 **4.4 Applications of laboratory results to the Earth's interior**

The comparison between laboratory data and seismological signals requires experiments in which the molten phase is in textural equilibrium with the solid matrix. Due to time-limited laboratory experiments, transient conditions may affect the results of acoustic velocities. In this case, textural analysis is important for correct interpretation of experimental data and run products. In a partially molten system at given pressure and temperature, the melt network can evolve to minimize the energy of melt-solid interfaces. This equilibration process concerns the

wetting angle θ at solid-solid-melt triple junctions, the area-to-volume ratio of melt pockets at 525 grain corners and the melt permeability threshold [Laporte et al., 1997]. The small dihedral 526 angles estimated for our partially molten samples ensures complete grain boundary wetting and 527 melt interconnectivity even for extremely low melt volume fractions [von Bargen and Waff, 528 1986; Laporte et al., 1997; Laporte and Provost, 2000], which is crucial for propagation of 529 seismic waves. The solid-melt dihedral angle is known to vary with pressure, temperature and 530 with the composition of the melt phase [Minarik and Watson, 1995; Yoshino et al., 2005]. 531 Experimental studies suggest that textural equilibration is a time-dependent process, which 532 533 usually requires long annealing times (weeks or months) [Waff and Blau, 1982; Laporte and *Provost*, 2000, *Maumus et al.*, 2005]. Still, small dihedral angle (10-30°), the extensive wetting 534 of crystal faces and the smoothly curved solid-melt interfaces observed in our samples are strong 535 indications that the microstructure has reached transient conditions and forming a melt solid 536 network close to equilibrium textures [Cooper and Kohlstedt, 1984; Waff and Faul, 1992; 537 *Cmíral et al.*, 1998] (Fig.4). Further, after reaching the peak temperature, the acoustic velocity 538 remains nearly constant (Fig. 2), suggesting that the samples are well relaxed, enabling a safe 539 comparison of our seismic wave velocity measurements with geophysical observations. 540

In addition, the extrapolation of laboratory acoustic wave velocities measurements to natural observations require the consideration of both anelasticity and frequency effects. Laboratory experiments (when not torsional) are usually performed at the frequency range from 20 to 50 MHz. The choice of this frequency range is determined by both requirements on excitation frequencies for the piezo-electric transducer as well with the restricted size of the probed samples in HP-HT apparatus.

Both anelasticity and anharmonicity that are accounting for the temperature dependence 547 of sound velocity could lower the observed velocities. These are functions of frequency, 548 temperature, pressure, mineral/melt intrinsic properties (including the chemical composition) as 549 well as grain-size and grain boundaries micro textures [Rivers and Carmichael, 1987; Karato, 550 1993; Jackson et al., 2002, 2004; Faul et al., 2004]. In solids, anharmonic effects related to 551 thermal expansion $(\partial \rho / \partial T)$ were found to be important in high frequency (MHz) experiments. 552 This process does not imply energy loss and remains nearly insensitive to frequency [Karato, 553 1993]. On the other hand, anelasticity is associated to energy loss and depends on frequency and 554 555 relaxation effects. Relaxation effects are thermally activated, hence anelasticity must be accounting for a significant part of the attenuation at high temperatures. Anelasticity of partially 556 molten system have been poorly studied [Faul et al., 2004; Jackson et al., 2004]. Because our 557 melt fractions are small (≤ 2 %) and similar to melt fraction estimated in the mantle, the 558 assumption of using anelastic values of pure olivine is reasonable [Jackson et al., 2002; Chantel 559 et al., 2016], also partially molten systems were found to have similar grain boundary sliding 560 process to solid [Jackson et al., 2002; Faul et al., 2004]. Nevertheless, this process is more easily 561 activated in melt bearing samples, where weaker grain boundaries have been reported [Faul et 562 563 al., 2004]. In addition, most of silicate melts have very high absorption in this frequency range and signals are seriously attenuated [Rivers and Carmichael, 1987]. However, this study showed 564 that the echo overlap technique is suitable for high-Q melts, and thus appropriate for MORB 565 566 melts. This study also stressed that for low viscosity melts (<1000 Pa.s), which is the case of MORB melts at high temperature (presence of volatiles will significantly increase this effects), 567 velocities are independent from frequency, as expected when wave's period is much smaller than 568 569 the characteristic relaxation time $(1/f \ll \tau)$. Relaxation time was estimated using the relation $\tau =$

570 $0.01*n*\beta$, where β is the inverse of adiabatic compressibility and n the melt viscosity, used for fitting of theoretical and experimental dispersion curves by Rivers and Carmichael, (1987). 571 Calculation using their parameters for the Kilauea basalt (1700 K) yields relaxation time of 0.467 572 nanoseconds for frequency of 30 MHz (used in our study). The product of angular frequency by 573 relaxation time between 10^{-2} and 10^{-1} (0.088) can be converted into a C/C₀ ratio (see fig 10 574 therein). It estimates the measured velocity to be similar to the relaxed one as the ratio is very 575 close to unity ($1 \le C/C_0 < 1.10$), pointing a very small effect of anelastic behavior. Our 576 moderately hydrous MORB probably have a lower viscosity and accordingly a shorter relaxation 577 578 time favoring our conclusions.

Detailed discussion on quality factor (Q) estimation by ultrasonic experiments, on similar compositions, has already been made by Chantel et al., (2016). However, our $\partial \ln(Vp)/\partial T$ and $\partial \ln(Vs)/\partial T$ values of -4.95 and -8.78 (*10⁻⁵) K⁻¹, of our olivine + MORB samples prior melting, are somewhat similar with temperature dependence values calculated by high Q from Karato, (1993), indicating a good agreement with pure olivine data up to melting point with anharmonic plus anelastic behavior.

Finally, the use of MHz frequencies tends to underestimate the effect of anelasticity. This increases our uncertainty on our measurements, but this uncertainty must be reasonable as shown by the small errors estimated in absorption calculations (4%) [*Rivers and Carmichael*, 1987], as well with near relaxed sound speed found for melt. In this study, we thus report a minimal effect of the presence of partial melt on the acoustic wave velocities and consider similar bias as estimated for solids (anelasticity and anharmonicity) because small fraction of melt seems to have only a moderate effect. Our extrapolation suffers also from grain size considerations as detailed by Jackson et al. (2002), even though this process was found to be nearly frequencyindependent for attenuation at mantle conditions.

594

595 **4.5 Geophysical implications**

Our study demonstrates that the melt content in a partially molten media can be better quantified by using the reduction of seismic velocity for melt fractions with a minimum detectable melt fraction between 0.5 % and 0.1 % (no seismic velocity drop seen for 0.1 %). On the other hand, for the studied melt fractions of 0.1-2.0 % with well-developed interconnectivity, electrical conductivity, varies within a strict range of about 0.5 log units even if transient values were only reached, too narrow to resolve fine melt structures without introducing significant uncertainties.

In this study, we specifically use the % drop in acoustic wave velocity as a measure to 603 determine the melt fraction. Our study indicates that the 3-8 % global reduction in seismic 604 velocity (Vs) observed at the top of the asthenosphere [Anderson and Sammis, 1970; Widmer et 605 al., 1991; Romanowicz, 1995] can be explained by 0.3-0.8 % volatile-bearing melt (Fig. 7a). 606 However, these values may vary laterally depending on the extent of melting at the 607 608 corresponding temperatures and the volatile contents in the mantle [Sifré et al., 2014]. Regional Vs variations of up to 10 % observed below the Pacific plate [Schmerr, 2012] indicate large melt 609 fractions of up to 1 % present in some parts of the asthenosphere, suggesting large 610 611 heterogeneities in terms of melt distribution. Apart from the global reduction of seismic velocity, numerous studies report velocity perturbations in various geological settings such as spreading 612 ridges, intraplate mantle plumes and subduction. [e.g. Pommier and Garnero, 2014]. Assuming 613 614 melt chemistry does not have any significant influence on acoustic wave velocity [Rivers and *Carmichael*, 1987]; we compare our melt fraction estimations to those reported using theoreticalmodels (Table S2).

In addition to the use of absolute sound wave velocities and velocities drop, the use of 617 618 Vp/Vs ratio can be of significant interest for comparison with seismological data. Bulk and shear 619 moduli of a partial melt system (a solid containing pore spaces saturated with melt) varies as a function of melt volume fraction. Accordingly, the relative change of Vp/Vs ratio could 620 621 indicate the presence of melt in deep mantle conditions. As detailed in Chantel et al., 2016, the absolute velocities values measured on analog systems do not compare well with real seismic 622 623 velocities measurements. It is mainly due to the difference in mineralogy (e.g pure olivine against peridotites) and relaxation effects due to differences in frequencies between natural and 624 experimental seismic velocities estimations (see discussion therein). However, the use of Vp/Vs 625 ratios and its variations allow a relevant comparison of our analog data to natural system as 626 627 changes are relative and not based on absolute velocities values.

Our data indicate that below melting point, Vp/Vs ratio increases from 1.75 to 1.8 from 628 room temperature up to 1650 K (Fig. 8). These values are consistent with values observed for 629 630 solid upper mantle ranging from 1.7 to 1.8 given by standard models such as PREM [Dziewonski and Anderson, 1981] or AK135 [Kennett et al., 1995]. These values are also consistent with 631 moderate Vp/Vs values obtained from upper mantle minerals, ranging also between 1.7 and 1.8 632 for olivine (1.8), clino and othropyroxenes (1.72 and 1.74) at the same pressures [Li and 633 Liebermann, 2007]. At melting, we observe a strong and sudden increase of the Vp/Vs ratio. The 634 magnitude of the increase correlates positively with the melt fraction. Vp/Vs ratios above 1.9 are 635 observed for sample with 2% melt fraction (Fig. 8 c). Our Vp/Vs ratio values compare favorably 636 with LVZ estimations with ratios given by global models ranging between 1.8 and 1.85 and 637

requiring only moderate amount of melt (<1 % of melt). However, our data implies very high melt fractions involved in local anomalies where Vp/Vs ratios up to 2.5 or more have been reported [*Schaeffer et al.*, 2010, *Hansen et al.*, 2012]. These very high anomalies imply higher melt fractions (12.5 % for Vp/Vs ratio of 2.5, based on the trend defined on Fig. 8 c) even if other physical processes such very high volatiles contents in melts could explain these anomalies.

In general, we find that the melt contents reported in previous geophysical studies are 644 consistently higher than the estimations based on laboratory measurements of seismic wave 645 velocities. The majority of these studies used the theoretical prediction of velocity reduction for 646 partially molten rocks, which underestimate the effect of melt fraction on seismic velocity. The 647 use of our laboratory measurements provides melt fractions that are consistent with petrological 648 models. Further, we believe that the refined melt fraction estimations would provide a solid 649 platform to constrain a meaningful cross correlation between field-based seismic and electrical 650 observations. The effect of the chemical composition of melt on acoustic wave velocity is one of 651 the important aspects worth exploring in future studies. 652

653

654 **5. Conclusions**

This study presents the first simultaneous measurements of electrical conductivity and acoustic wave velocity of partially molten samples of geophysical importance. The results highlight how electrical conductivity and acoustic wave velocity respond to the evolving melt texture from a completely random melt distribution. The continuous increase of electrical conductivity at constant temperature, after melting of MORB, indicates that the melt interconnectivity evolves with time. In contrast, constant seismic velocity after the melting

suggests acoustic velocity is sensitive to the melt volume fraction in the sample, but less affected 661 by the evolving melt texture. Our results suggest that the electrical conductivity of partially 662 molten materials measured before reaching the evolved melt interconnectivity can lead to an 663 underestimation of the EC for a given melt fraction. This may result in an over-estimation of 664 melt fraction in geological settings. Overall, the Vs measurements appear to be a more 665 666 appropriate method for determining the melt fraction in a partially molten system with complete wetting properties. The previous approximations based on theoretical models of seismic velocity 667 appear to overestimate the extent of melting in the mantle. This study demonstrates the necessity 668 669 of using electrical conductivity values from texturally equilibrated partially molten sample for comparison with geophysical data. 670

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931 Figures and Table



Figure 1. Schematic cross section of the high pressure cell assembly for simultaneous acoustic velocity and electrical conductivity measurements. The assembly was designed in order to place the sample at the centre of the cell at high pressure due to differential compression of alumina and MgO pistons. The respective pistons lengths were calibrated in prior experiments.



Figure 2. (a) *S*- and (b) *P*- wave velocities as a function of increasing temperature (left) and as a function of time at a constant temperature (right) for olivine- and melt-bearing samples investigated in this study. The uncertainty results from the estimations of temperature, pressure, sample dimensions and data fitting errors and are estimated to be lower 2.6 % (2σ) of the value. Errors are not represented for visibility. Temperature error (10 K (2σ)) is within the symbol for all data points.



Figure 3. Electrical conductivity as a function of reciprocal temperature (left) and, as a function of time at a constant temperature (right), for the samples containing 0.1, 0.5 and 2 vol. % melt fractions. Solid and dashed lines indicate the corresponding conductivities immediately after melting and after 1 hour at constant temperature of 1650 K. The uncertainties associated with the electrical conductivity data measurements are less at high temperatures. The uncertainty results from the estimations of temperature, pressure, sample dimensions and data fitting errors and are less than for EC 5 % (2σ) of the value. Error in temperature is 10 K (2σ).



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Figure 4. Backscattered electron (BSE) images showing equilibrium melt textures at meltolivine interfaces at 1650 K. Well-developed interconnected melt channels in samples with (**a**, **b**)

955 2.0 %, (c, d) 1.0 %, (e, f) 0.5 % and (g, h) 0.1 % melt volume fractions. The bright particles 956 attached on sample surface and holes are rhenium pellets coming from polishing (TC's, furnace 957 and electrodes are made of Re). Biggest particles are presents in samples holes and along 958 sample/capsule interface (4c and 4g) whereas smallest particles are attached to sample surface 959 along polishing scratches (well visible in 4f).



Figure 5. Comparison of reported acoustic velocities and electrical conductivities for partially 961 molten systems. (a) Vp,s /Vp₀,s₀ ratios for various melt fractions. Experimentally determined 962 963 velocities for the olivine-basalt system [Chantel et al., 2016] and theoretical estimations for olivinebasalt [Yoshino et al., 2005] and melt analogue systems [Takei, 2000] are shown for comparison. 964 Errors in SV ratio are 3.6 % (2σ), errors in melt fraction are within the data symbol (1% relative). 965 966 (b) Reported electrical conductivity values for olivine, MORB and olivine + MORB compositions. Open circles and filled circles indicate our conductivity data for our volatile-bearing partial melts at 967 melting and data after 1h a 1650K. The conductivity values presented in the figure are C06 968 [Constable, 2006], P72 [Presnall et al., 1972], TW83 [Tyburczy and Waff, 1983], N11 [Ni et al., 969 2011], Y10 [Yoshino et al., 2010], Z14 [Zhang et al., 2014], C11 [Caricchi et al., 2011], M05 970

971 [*Maumus et al.*, 2005] and L17 [*Laumonier et al.*, 2017]. The EC values reported in [*Zhang et al.*, 972 2014] indicate melt conductivity before (open circle) and after (solid circle) the textural 973 modifications due to the shear deformation. Errors on our data points are 5 % (2σ) on EC value and 974 10 K (2σ) in temperature.



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Figure 6. The comparisons of electrical conductivity before and after 1 h at 1650 K for studied 976 977 melt fractions. The conductivity corresponding to the LVZ is shown by the blue shaded area. The vertical lines indicate the minimum melt fractions required to explain the high conductivity zone 978 in the asthenosphere. Extrapolated data using the logarithmic laws fitted from data in figure 3 979 980 were represented from relevant time-scales with squares. This slow increase shows that weeks of equilibration will be necessary to have half an order of magnitude increase of EC. Such an 981 increase is compatible with measurements performed on equilibrated material after weeks of 982 annealing. Errors on EC are 5 % (2σ) and within the data symbol for melt fraction (1 % relative). 983 Errors on extrapolated values are 10 % (2σ) and we not displayed for distinction with measured 984 data points. 985



Figure 7. (a) The % drop in P- and S-wave velocity as a function of the sample melt fraction. 988 The geophysically observed S- wave velocity anomaly for the LVZ in the asthenosphere is 989 shown by the shaded rectangle. Note that the 3-8 % Vs drop observed for most of the 990 991 asthenosphere can be explained by 0.3-0.8 % melt. (b) Vs velocity anomalies observed at various geological settings and the possible melt fraction estimations based on our model. The Vs 992 anomalies presented in the figure are from EPR 17S [Toomey et al., 1998], EPR 8-11N [Toomey et 993 al., 2007], Yellowstone-Snake River [Wagner et al., 2010], Hawaii [Laske et al., 2011], and the 994 Pacific [Kawakatsu et al., 2009; Schmerr, 2012]. Errors on velocity drops are 5 % (25) relative and 995 within the symbol for melt fraction (1 % relative). 996



Figure 8. (a) Vp/Vs ratio of our experiments reported as a function of increasing temperature. 998 The standard deviation on the data lies between 0.305 and 0.335 (lowest to highest temperature 999 1000 values) corresponding to an error majored by 1.75% of the Vp/Vs ratio (1 σ). b) Vp/Vs ratio as a 1001 function of the volume melt fraction (MORB content). Values for solid sample before melting are represented by circles and values after partial melting occurred with squares. Vp/Vs ratios from 1002 seismological data: PREM [Dziewonski and Anderson, 1981] and AK135 [Kennett et al., 1995] 1003 1004 are represented in black dashed lines. Vp/Vs ratios from nominal minerals are represented by colored zones (green for Olivine, grey for OPX and gold for CPX), velocities data at 2.5 GPa 1005 were taken from [Li and Liebermann, 2007]. c) Inset figure in a) Vp/Vs ratio increase at melting 1006 in function of melt fraction, quantifying the magnitude of increase of the ratio in response to 1007 partial melting. 1008

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	MORB	MORB	Olivine	Olivine
	initial	final	initial	final
SiO ₂	50.73 (0.28)	51.15 (0.1)	41.97 (0.8)	41.43 (0.4)
TiO ₂	2.00 (0.02)	2.22 (0.27)	0.06 (0.02)	0.03 (0.02)
Al ₂ O ₃	13.73 (0.18)	14.54 (0.1)	0.1 (0.03)	0.1 (0.01)
FeO	11.29 (0.13)	9.20 (0.1)	9.3 (0.3)	9.15 (0.2)
MnO	0.2 (0.2)	0.22 (0.1	0.11 (0.03)	0.12 (0.04)
MgO	7.1 (0.12)	6.94 (0.09)	48.65 (0.9)	49.69 (0.7)
CaO	10.94 (0.22)	11.41 (0.03)	0.23 (0.1)	0.22 (0.01)
Na ₂ O	2.83 (0.07)	2.26 (0.3)	< 0.01	< 0.01
K ₂ O	0.15 (0.08)	0.35 (0.1)	< 0.01	< 0.01
Total	98.79 (0.5)	98.29 (0.2)	100.4 (0.2)	100.7 (0.8)

Table 1. Chemical composition of MORB and olivine before and after the experiments.